

Decarbonizing Commercial and Industrial Heating with Hydrogen

Part 1 – Fundamentals, Materials Impacts, and Commercial Cooking/HVAC Test Results

Paul Glanville, P.E., **GTI Energy**

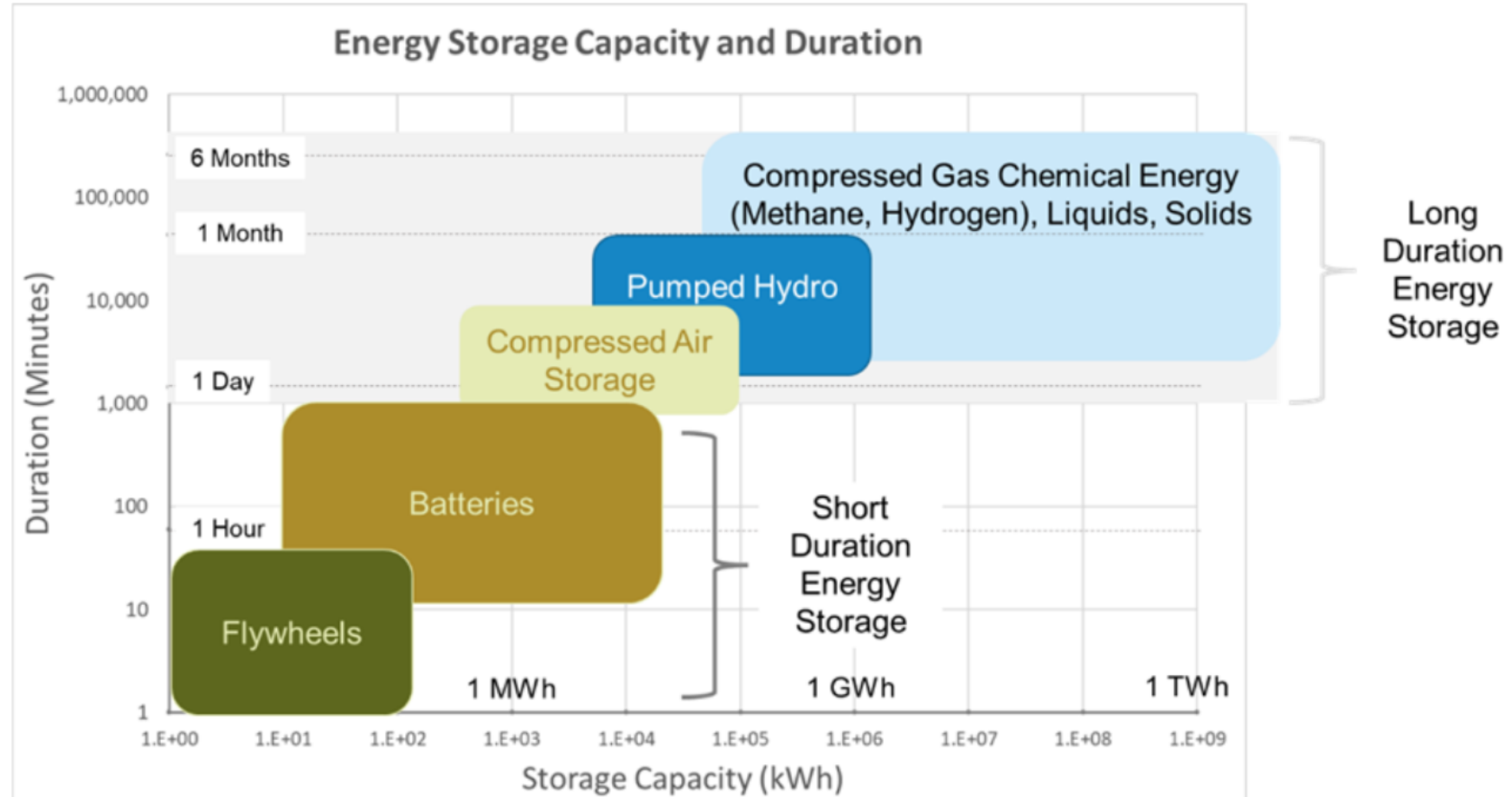
Prof. Vince McDonell, **Univ. of California Irvine**

February 11, 2025

Decarbonize our Largest Energy Storage Asset

North American Gas Infrastructure

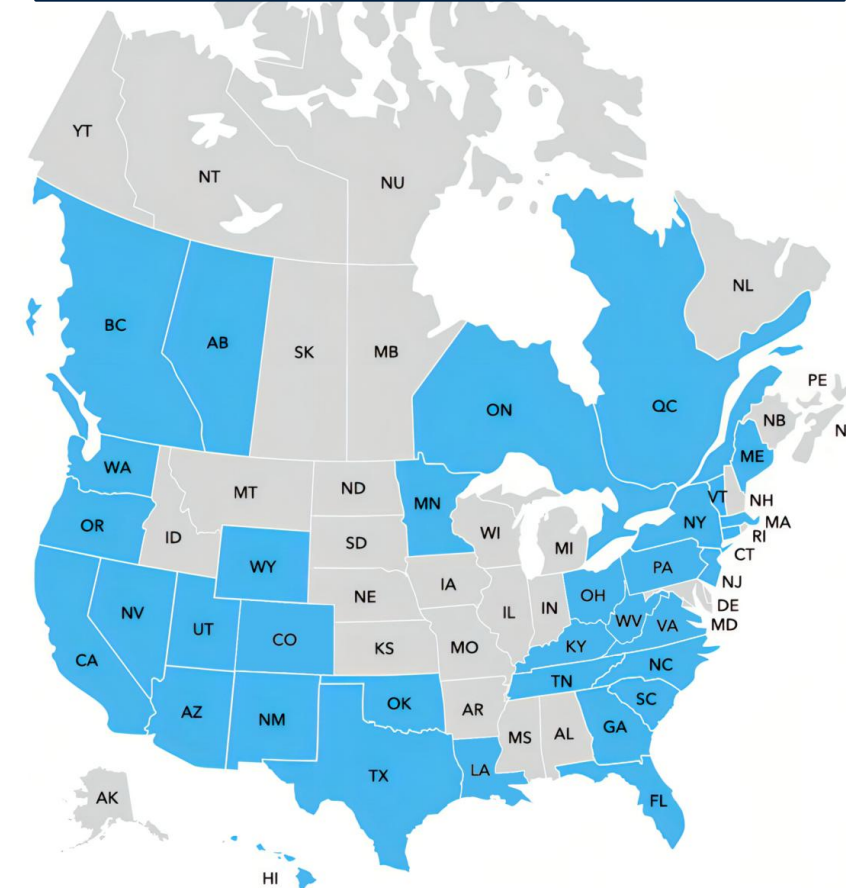
- Spans 5.4e6+ km
- Serves 85+ million buildings
- Delivers 600+ GW of peak energy
- Stores 1.3+ PWh over months
- Results in >1,000 MMT CO₂/y (heat)
- Can play a constructive and resilient role in **decarbonization**, in transition and in the future



Hydrocarbons: Drop-in for Natural Gas/Propane with *Net* GHG reductions

- ## Hydrogen: **Not a Drop-in** but with *Absolute* GHG Reductions

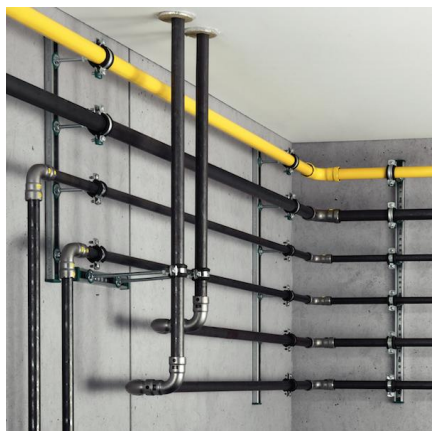
- ## States/Provinces with 1+ Active H₂ Blending/Distribution Projects



Hydrogen Economy – The Industry is Moving Fast!



Gas Train: Full product line OK for 25% H₂ today, 100% H₂ available



Customer Distribution: System options for 0-100% H₂ available today



Customer Distribution

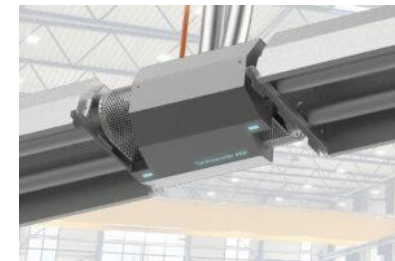
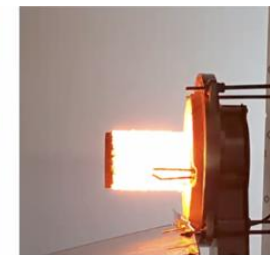
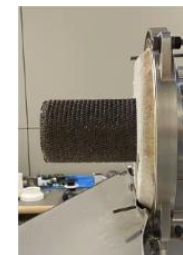
Heating Industry is Increasingly H₂-Ready

Product / Technology Demonstrations



Combustion Controls and Sensors: Full product line OK for 30% H₂ today, 100% H₂ available in a range of applications

Burners: Most vendors have burners certified with G222 (23% H₂), many have developed 100% solutions, with industry guidance on designs widely available.



Equipment: Some have 'self-certified' to blends, 20%-30% H₂ in NA (cert. possible now), nearly all end use categories have 100% H₂ developed for sale/demo

Gas Train

Combustion Controls

Burners and Equipment

Project Overview – PIR-22-001

California-Focused Project

Large effort to quantify the potential of hydrogen to decarbonize **large buildings and industry in California**:

- Develop techno-economic roadmap to decarbonize ~**50%** of CA's nat. gas use
- Large effort across diverse team to:
 - Develop CA-specific TEA for H₂ use, quantify potential/costs of conversions to H₂
 - Test/model H₂ tolerance of wide range of large equipment categories (e.g. boilers)
 - Material testing for long-term impacts
 - Air Quality simulation on regional impacts
 - Stakeholder outreach and engagement

Decarbonizing Large Commercial and Industrial Equipment with Hydrogen (PIR-22-001)



CALIFORNIA
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GTI ENERGY



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EPRI
ELECTRIC POWER
RESEARCH INSTITUTE

AHRI
AIR-CONDITIONING, HEATING,
& REFRIGERATION INSTITUTE



Pacific Gas and
Electric Company



SOUTHWEST GAS



UTD
Utilization Technology
Development



LCRI
LOW-CARBON
RESOURCES INITIATIVE

Test Equipment Selection

- 2+ units per equip. category
- Finalized after Preliminary TEA

Commercial Examples:

- Furnaces/Weatherized HVAC
- Water Heater/Hot Water Boilers
- Cooking / Catering Equipment

Industrial Examples:

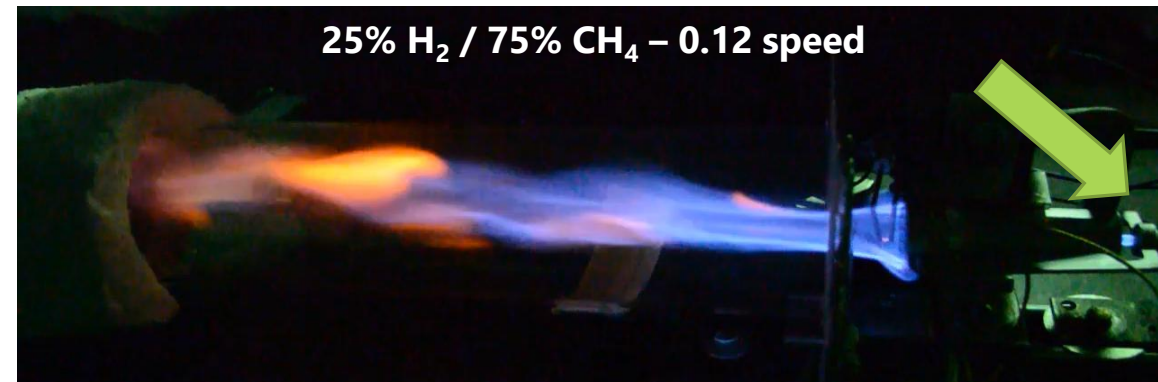
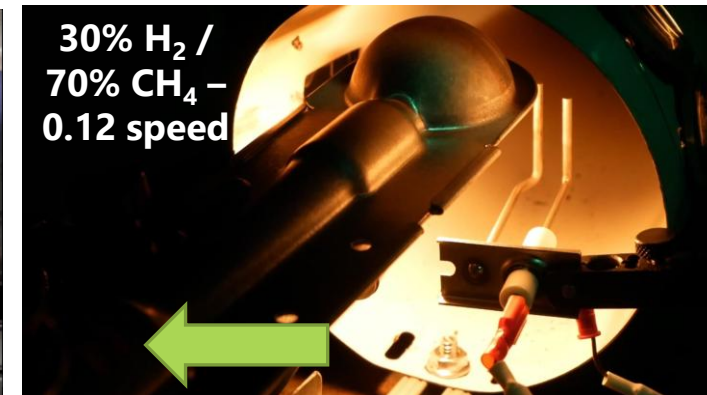
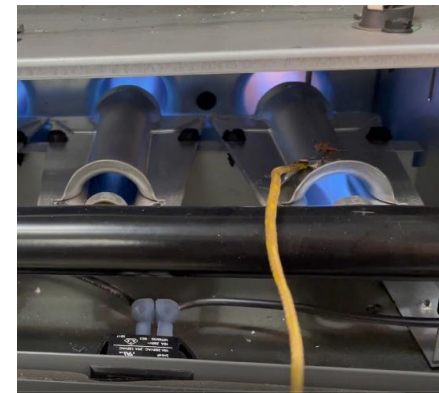
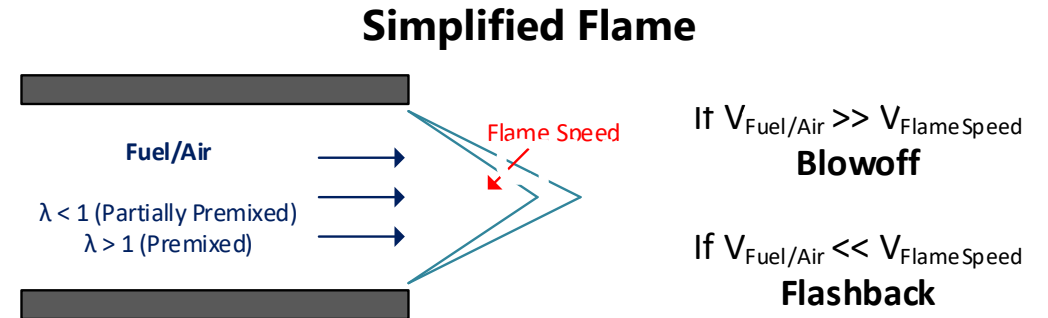
- Steam Boilers / Process Heaters
- Ovens / Dryers / Kilns
- Heat Treating / Furnaces



Project Motivation – PIR-22-001

What do we typically look for with H₂?

- Short-duration testing looks at...
 - Flame stability/safe ignition/flashback
 - Surface temperatures/Radiant Output
 - Capacity/Efficiency/Modulation
 - Emissions (NO_x, CO, CH₄, or H₂)
 - Impact of variable blending/balance fuel
 - Static leakage enhancement
- But what about...
 - Higher blends/pure hydrogen? Long-term impacts? Testing to failure?
 - Broader population of equipment (type, age, installation)? Emerging technologies and retrofit packages? Impact on industrial processes?



First Goal – Establish the Decarb. Potential

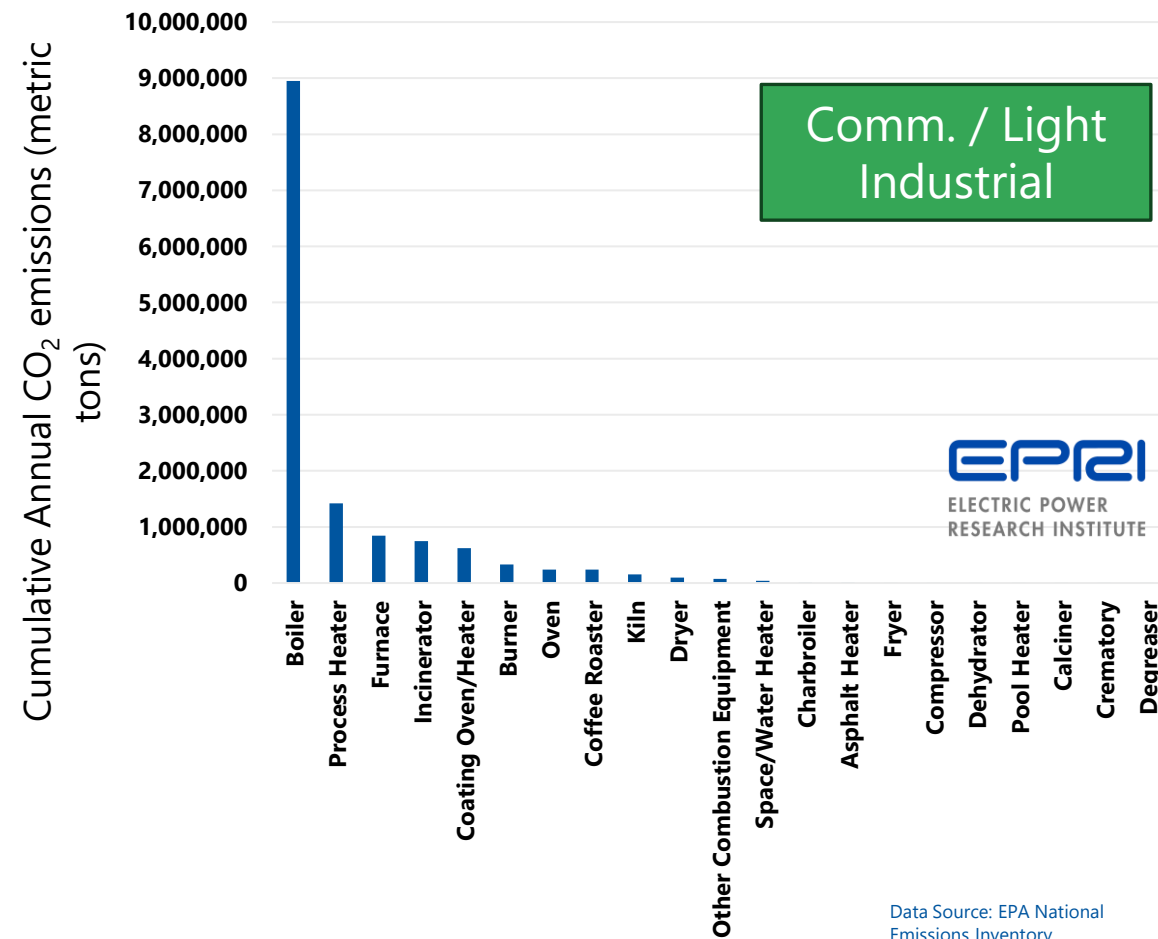
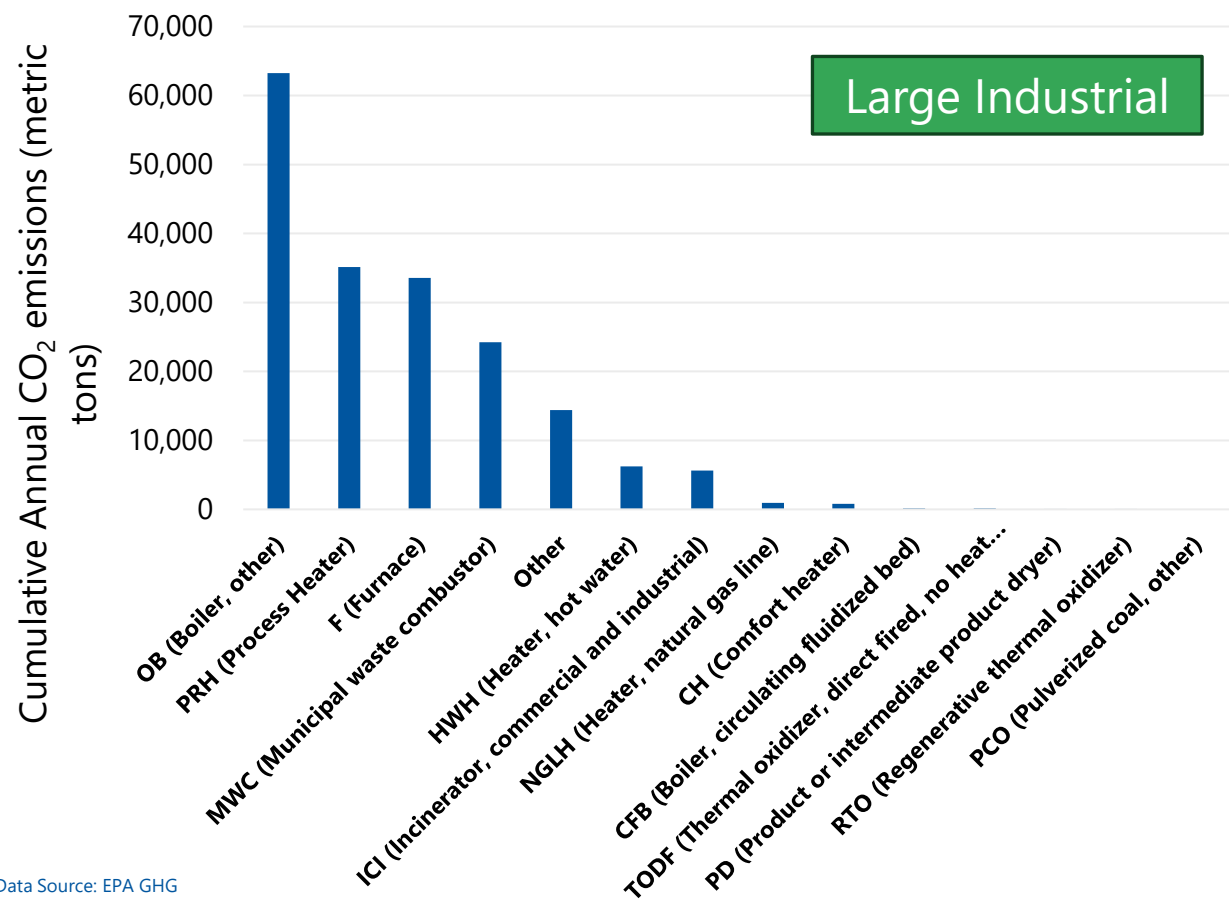
Goal of the task-level effort was to recommend equipment categories/applications for further investigation **based on GHG reduction potential with H₂-based fuels**

- Establish a CA-specific *Combustion Equipment Database* for TEA and broader project
 - Draft based on EPA, AQMD/APCD sources, continuing to seek data from CA IOUs and industry
- *Preliminary Techno-Economic Assessment* provides GHG potential of adapting H₂-based fuels in CA C&I sectors
 - EPRI model built and calibrated to DOE/CA databases (e.g. CBECS)
 - Using prior database and range of decarbonization scenarios, quantify energy/cost/emissions for:
 - **Natural Gas:** Reference Case, Maximum Achievable Energy Efficiency (MAEE)
 - **Low Carbon Fuels:** Blended Case (NG / RNG / H₂), Blended Case w/ MAEE, 100% H₂ w/ MAEE
 - **Alternative:** Partial Electrification (w/ LCFs), Widespread Electrification

Results discussed in 2nd session

Preliminary Results – Database (Comm. / Industrial)

Draft/Preliminary Results: Subject to Change



Boilers, Process Heaters and Furnaces have highest cumulative CO₂ emissions and highest equipment counts

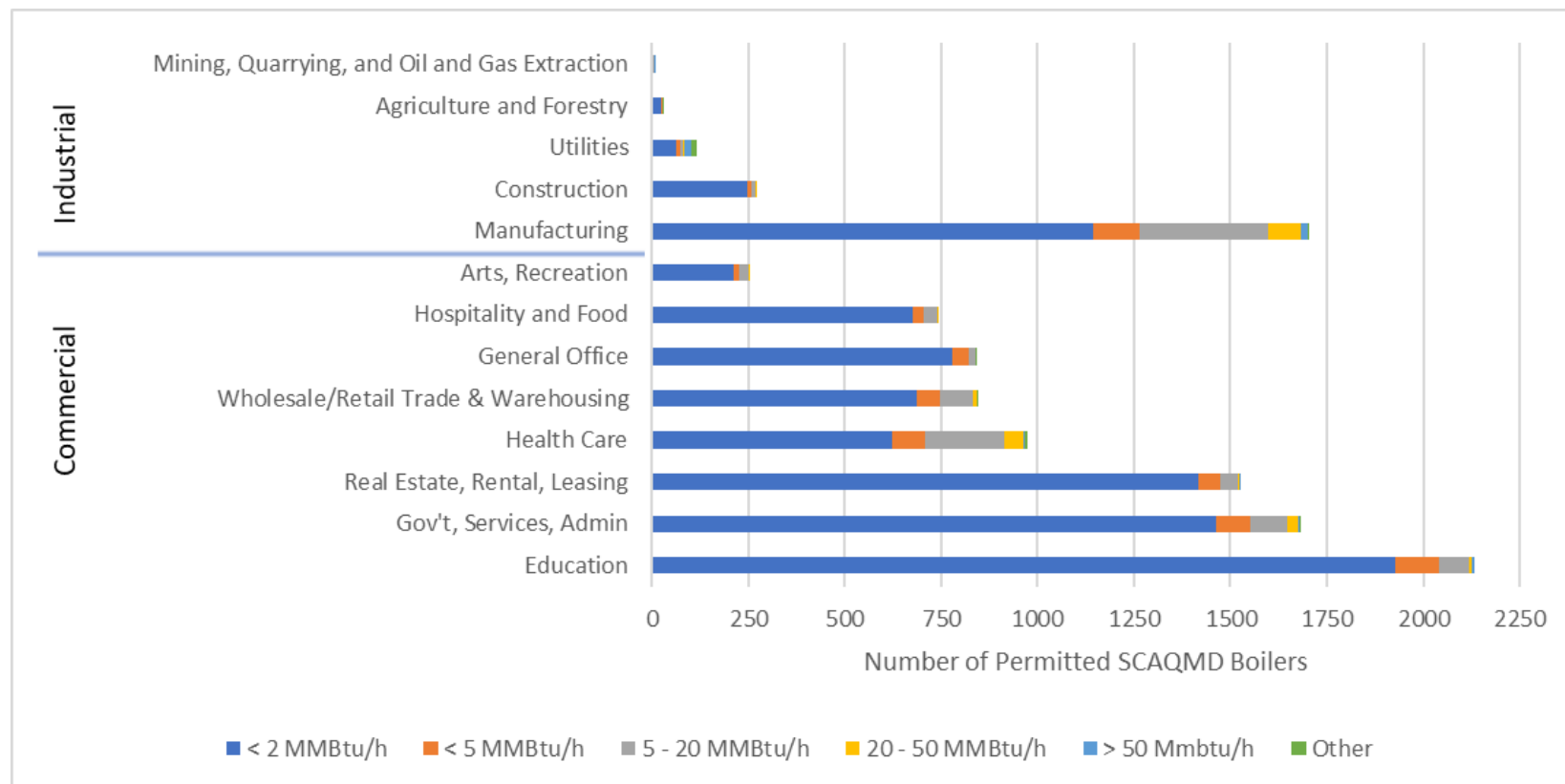
Preliminary Results – Database (Continuation)

Draft/Preliminary Results: Subject to Change

Project team sought further granularity/detail through local agency, utility, and industry group data sources, below is AQMD example to emphasize the importance of **boilers overall**

There are **even more** boilers beyond EPA permitting, typ. OEM-certified, more than 10X this estimate. Within the SCAQMD:

- 97% (10,800) are ≤ 20 MMBtu/h
- 99% are ≤ 50 MMBtu/h
- The combined energy demand of all these boilers ≤ 50 MMBtu/h is ~15 GW - Appx. 1/3 of the 2021 CAISO peak.



Data Source: SCAQMD

Second Goal – Filling the Gaps in the Data

Recommend equipment categories/applications for further investigation **based H₂ utilization potential and data/knowledge gaps**

- ***Technical Survey of Performance, Safety, and Emissions Impacts***

- Review the efficiency, operability, emissions, and safety impacts of combustion equipment operating with H₂-blended natural gas and, if suitable, 100% hydrogen
- Survey equipment/operational modifications to **tolerate higher blends of hydrogen**
- Identify research and technology gaps, concerning hydrogen utilization in key equipment categories/applications, **provide basis for test planning**

- Scoping of survey and recommendations for testing program:

- Large HVAC* equipment (commercial warm-air furnaces, rooftop units, unit heaters), Commercial cooking equipment – **Covered in Session 1**
- Industrial furnaces, Process ovens and dryers, other Industrial stationary combustion equipment, Boilers (hot water and steam), Water Heating - **Covered in Session 2**

*Per the CEC, "Large Commercial Building" is defined as a non-residential building with 100,000 ft² of floor space or greater.

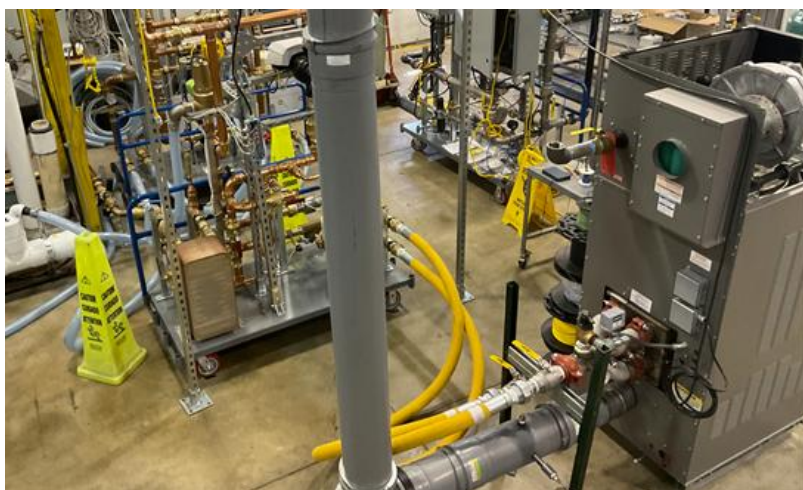
Testing Program – Filling the Gaps in the Data

Testing and Analysis Program:

- Test rigs for **six categories** of large commercial/industrial heating equipment underway now
- Examples of natural gas equipment tested with increasing hydrogen **two ways (on / off rate)** over 2024
 - Data collected on perf., emissions (NO_x , CO, **CH₄**, **H₂**), noise, etc.
 - Evaluate retrofit options for higher H₂
- Calibrate **CFD combustion model** for extrapolation to equipment/designs
- Investigate **impact on materials** (e.g. refractory) in parallel

Equipment Type	Sub-type(s)	Coverage Range	Test Unit Range
Boilers	Steam	Up to 50 MMBtu/h input	300 to 3,000 kBtu/h input
	Hydronic/Hot Water		
Direct-fired Process Heating	Ovens, kilns, and dryers	Up to 100 MMBtu/h input	500 to 2000 kBtu/h input
Industrial Furnaces	Recuperative / Non-recuperative Burners	Up to 100 MMBtu/h input	500 to 2000 kBtu/h input (200 to 500 kBtu/h Radiant tube)
Commercial HVAC	Warm-air Furnace, Duct Furnaces, & Unit Heaters	200 to 1,000 kBtu/h	
Commercial Cooking	A range equip.: fryers, broilers, griddles, ovens, charbroilers, and ranges	100 to 500 kBtu/h	

Increasing GHG Impact in CA



Decarbonizing Commercial and Industrial Heating with H₂

Session #1 (1:30p-2:30p)

"Hydrogen 101" (UCI)

- Review the fundamentals of hydrogen combustion relative to conventional fuels
- Understand the potential impacts on a variety of burner and combustion system designs

Research Project Plan & Results (UCI)

- Discuss the potential short/long-term impacts on materials within heating equipment
- Review the experimental test plan and preliminary results for Commercial Cooking and Commercial HVAC equipment

Session #2 (3:00p-4:00p)

Research Project Plan & Results – Cont. (GTI)

- Review the experimental test plan and preliminary results for Industrial Combustion Equipment, Boilers, and Water Heaters

Hydrogen – OEM Perspective (A.O. Smith)

- Manufacturer perspective on H₂ applied to heating equipment and testing results

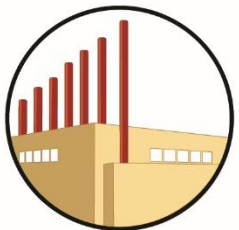
Pulling it All Together – H₂ Big Picture (GTI)

- Putting research data into broader context, including techno-economics, codes & standards, trends in test data, and H₂ safety



AHR Session

Decarbonizing Commercial and Industrial Heating with Hydrogen



**UCI Combustion
Laboratory**

UCIrvine | UNIVERSITY
OF CALIFORNIA

Vincent McDonell

“H₂ 101” - Fundamentals

- Hydrogen differs from the natural gas we are used to working with

	Methane	30% Hydrogen 70% Methane	Hydrogen
Molecular Weight (g/mol)	16	11.8	2
Density (lbm/ft³)	0.042	0.031	0.0053
Higher Heating Value (Btu/ft³)	1014	806	325
Wobbe Index (Btu/ft³)	1361	1261	1232
Stoichiometric Air-fuel Ratio (AF_{st}) (v/v)	9.5	7.4	2.4
Lower Flammability Limit (LFL) (% vol in air)	5%	4.7% ^a	4%
Upper Flammability Limit (UFL) (% vol in air)	14%	19% ^a	75%
Quenching Distance ($\lambda=1$) (in)	0.079 ^b	0.069 ^b	0.025 ^b
Minimum Ignition Energy in Air (10⁻⁵ J)	20 ^c	10 ^c	2 ^c
Adiabatic Flame Temperature ($\lambda=1$) (°F)	3542	3575	3807
Laminar Flame Speed ($\lambda=1$) (ft/min)	75 ^d	94 ^d	463 ^d
Min. Detonation Cell Size ($\lambda=1$, 68°F) (in)	12 ^e	7 ^e	0.4 ^e

a – Based on Le Chatelier’s rule [2]

b – Fukuda et al [3]

c – Hankinson et al [4]

d – Calculated using GRI-Mech 3.0 [5] and Cantera [6]

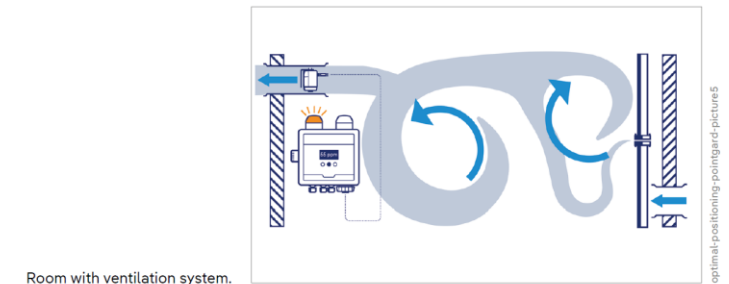
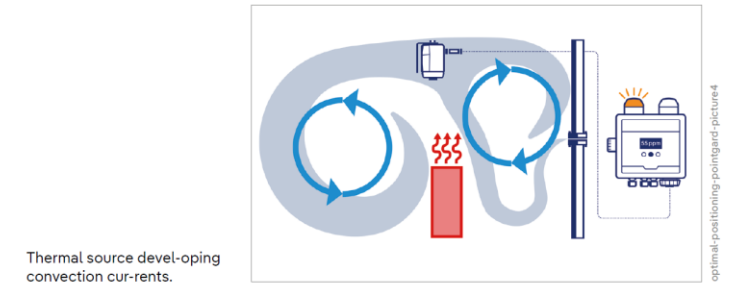
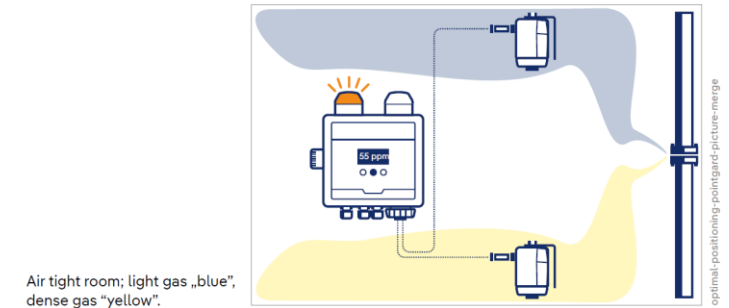
e – Matignon et al [7]



"H₂ 101" - Fundamentals

- Implications/Questions regarding density/MW
 - Dispersion, segregation
 - Leaks/accumulation
 - ✓ hydrogen can gather near ceilings of enclosed spaces, etc
 - ✓ Combined with range of flammability, considerations for ventilation/sensor locations
 - Will H₂/NG stratify in pipelines?
 - ✓ Molecular diffusion suggests no
 - Gases are not like liquids (oil / vinegar)
 - ✓ Empirical evidence
 - Analyzer span gases
 - Several recent studies
 - Will H₂ leak preferentially from pipes, etc?
 - Will H₂ interact with metals?

Different scenarios which are typical for gas leaks within a building



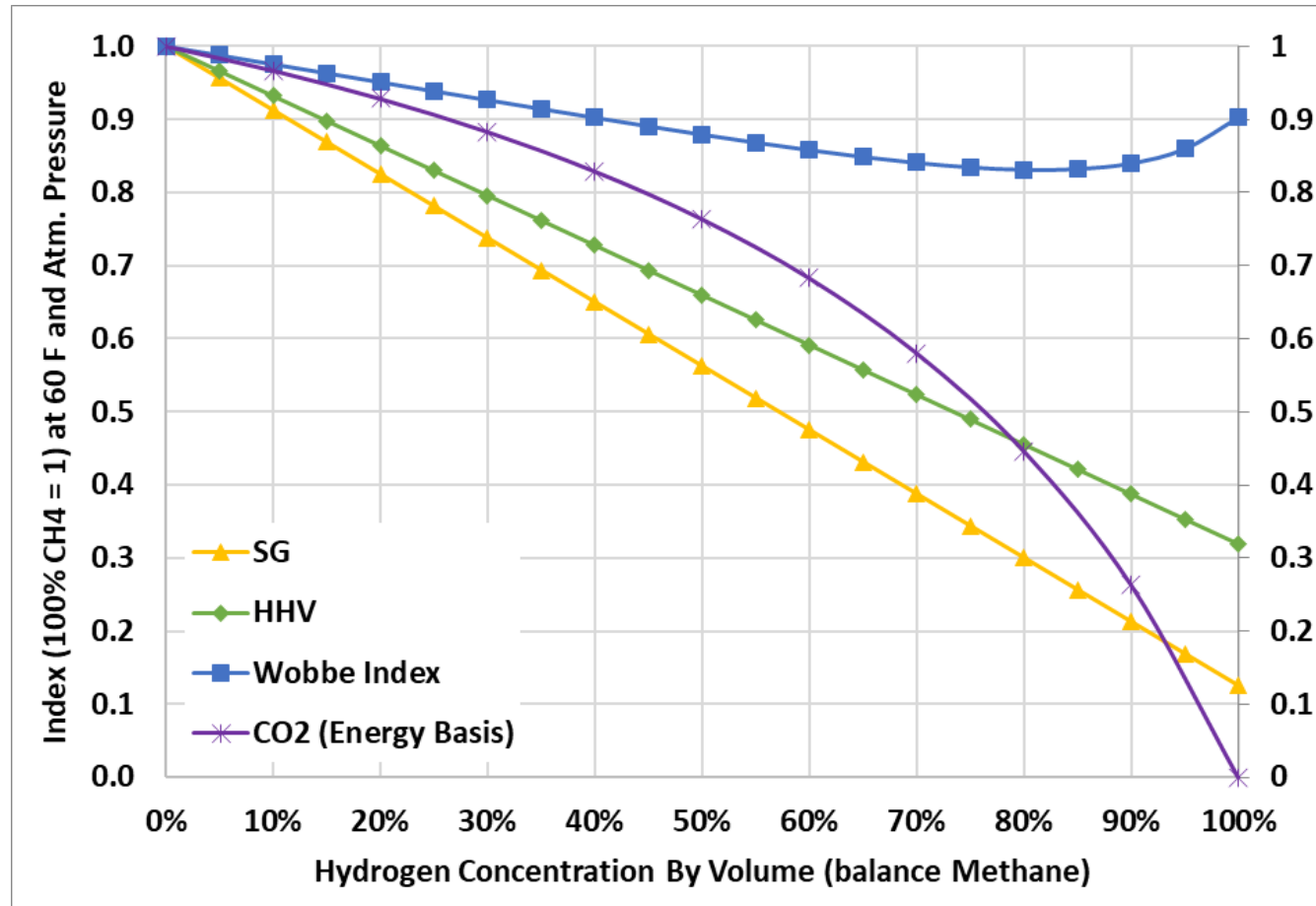
Source: Dräger (2008), Gas Dispersion, STL-1168



"H₂ 101" - Fundamentals

- Interchangeability

- Hydrogen is relatively interchangeable with natural gas (Wobbe Index)
- Must flow 3x volume to match gaseous heat input



Source: GTI Energy

$$Wobbe = \frac{HHV}{\sqrt{S.G.}}$$

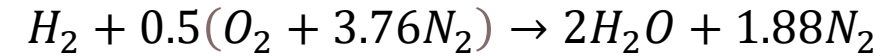
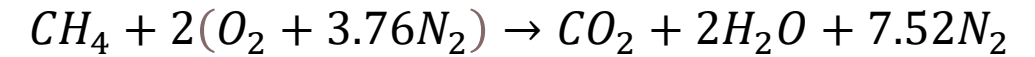
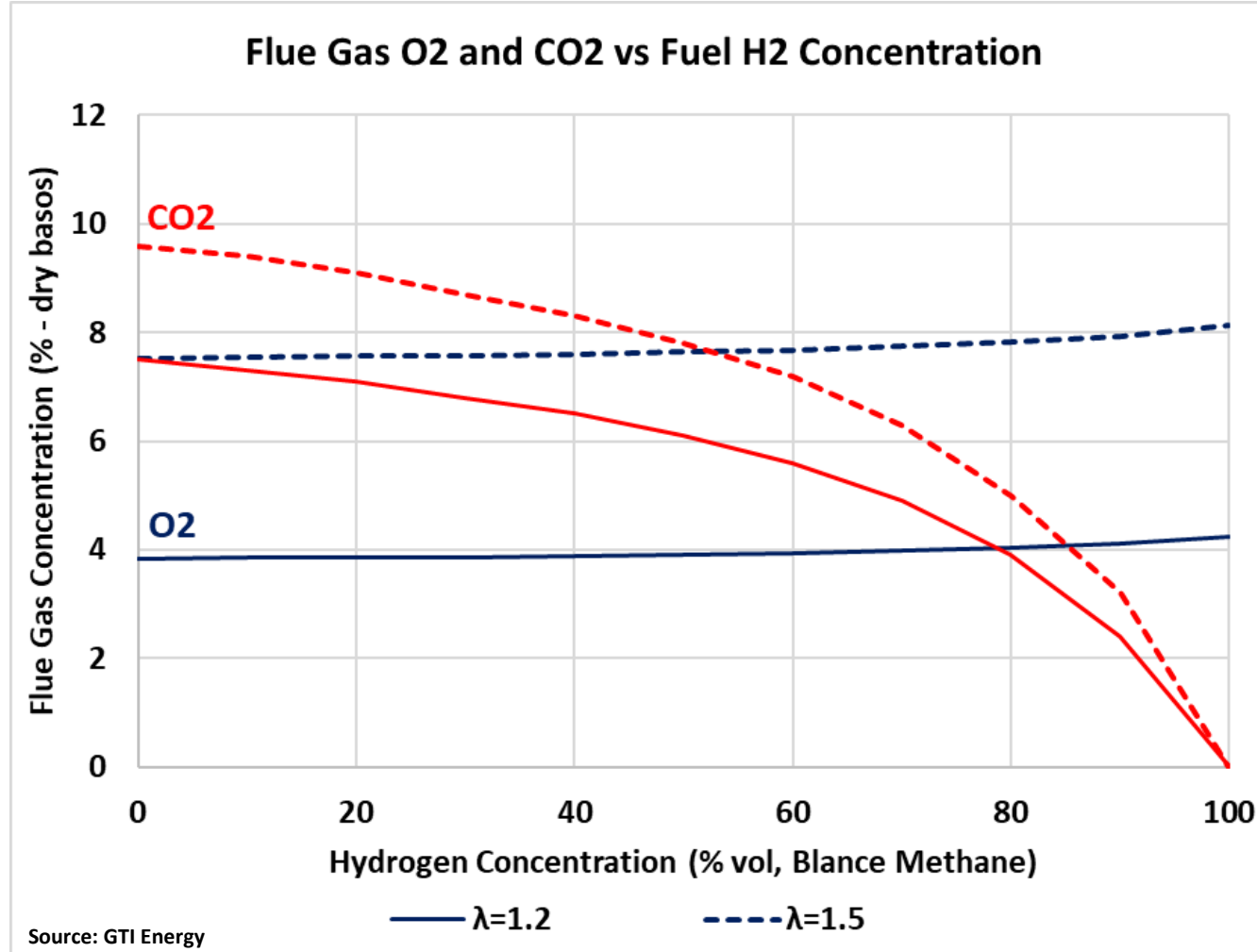
- How many BTUs can I get through a given orifice diameter with a given feed pressure?



"H₂ 101" - Fundamentals

- Air to Fuel Requirements

- Hydrogen requires about 24% less air to fully react than does natural gas



0.5 parts O₂ vs 2 parts O₂

But need ~3x more volume to
match heat input

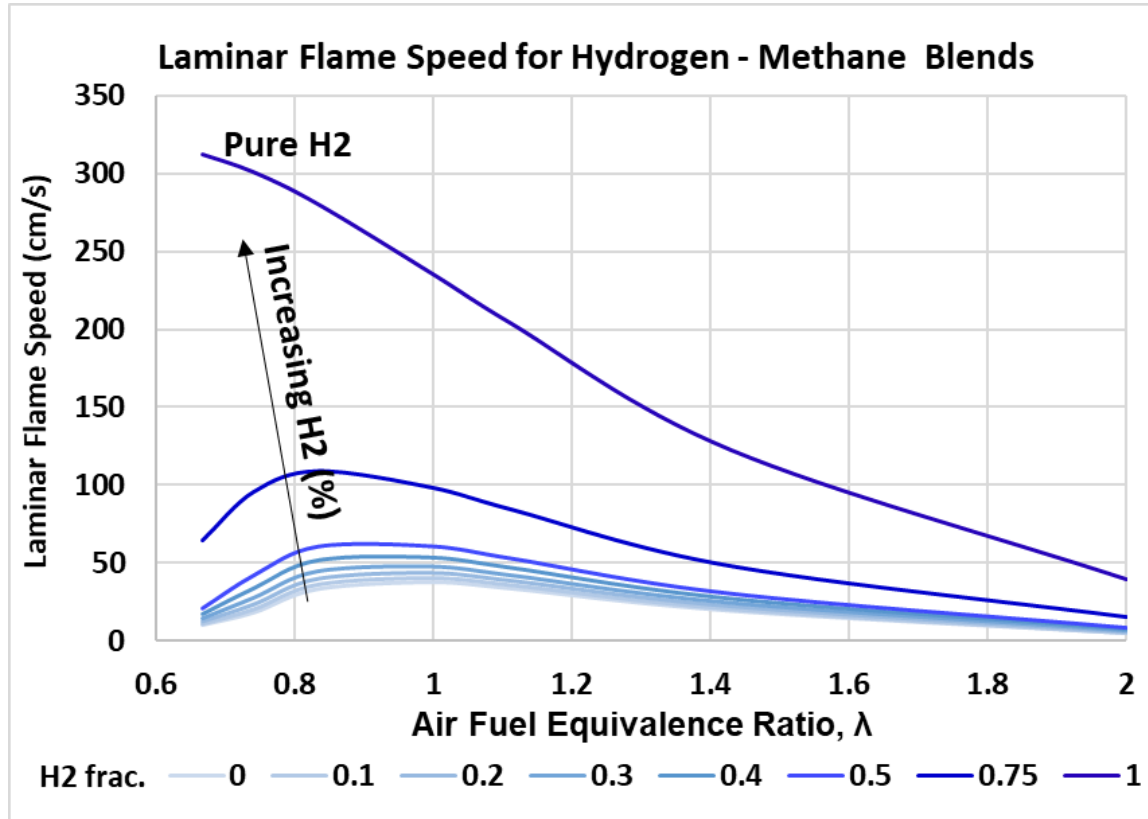
$$1014 \text{ BTU/scf} / 325 \text{ BTU/scf} = 3.12$$

$$\frac{\dot{V}_{air,H_2,st}}{\dot{V}_{air,CH_4,st}} = \frac{HHV_{CH_4}}{HHV_{H_2}} * \frac{0.5}{2} = 3.12 * 0.25 \approx 0.76 < 1$$

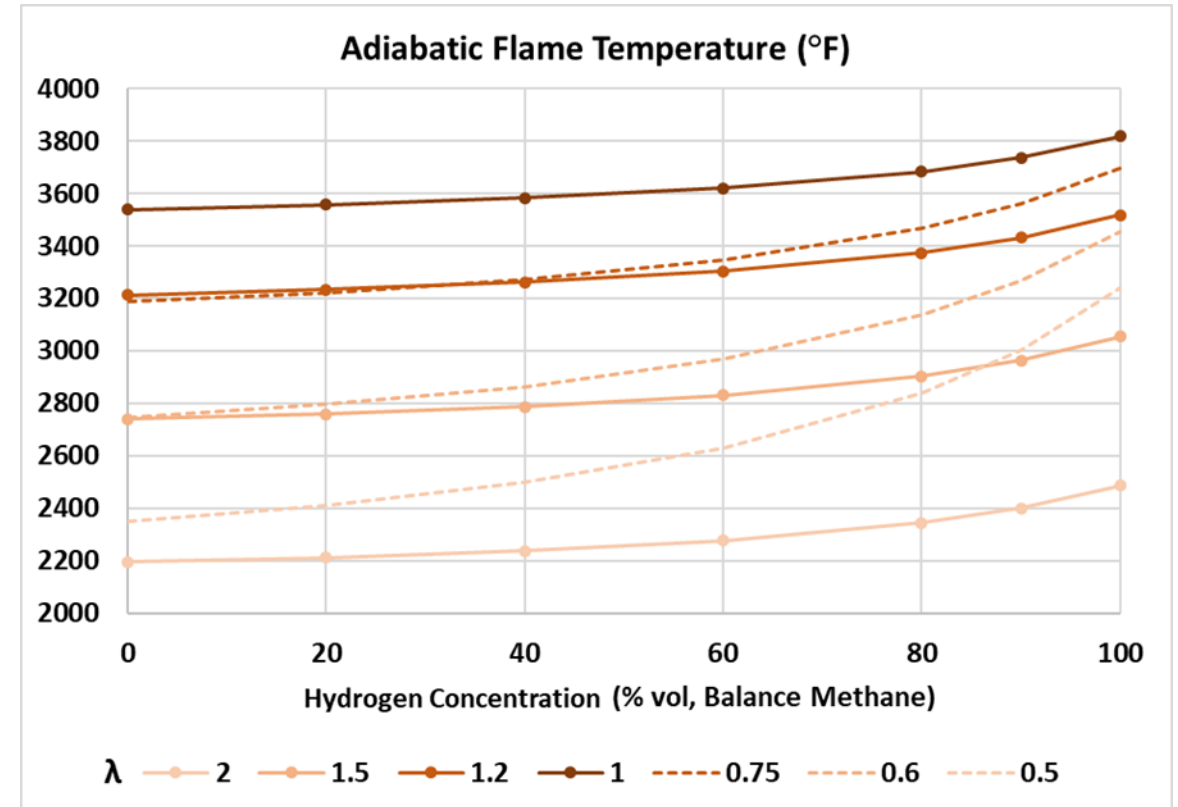


"H₂ 101" - Fundamentals

- Flame Speed and Flame Temperature
 - Hydrogen burns faster and hotter than natural gas (but.....)



Source: GTI Energy

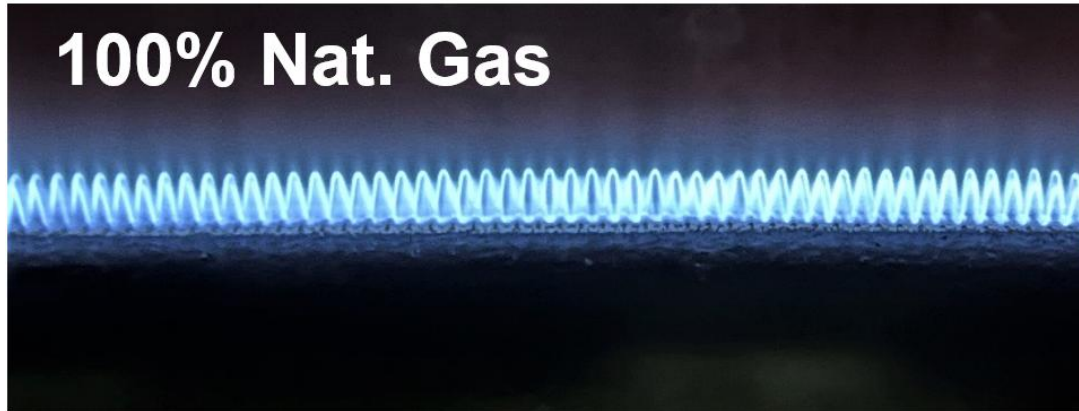


Source: GTI Energy



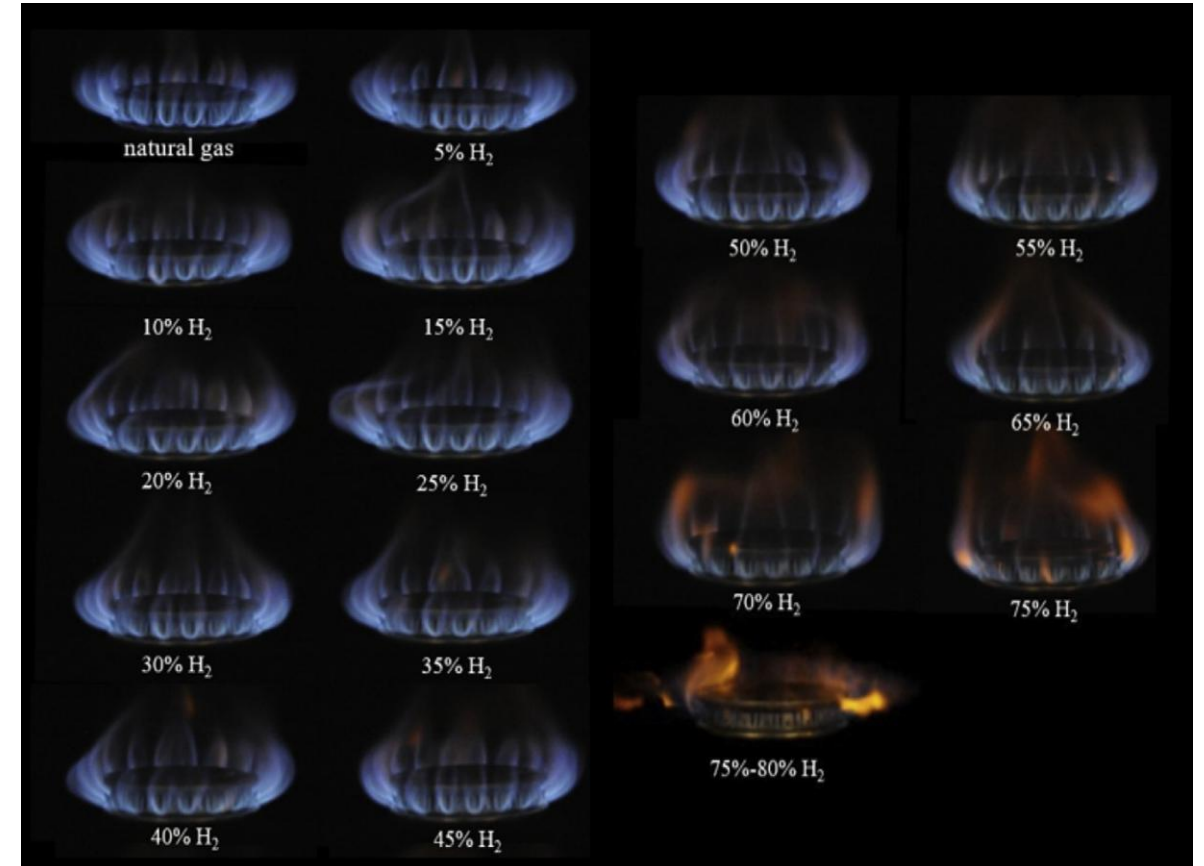
"H₂ 101" - Fundamentals

- Luminosity differs



Source: GTI Energy

Hydrogen: colorless, odorless....
What about heat transfer?

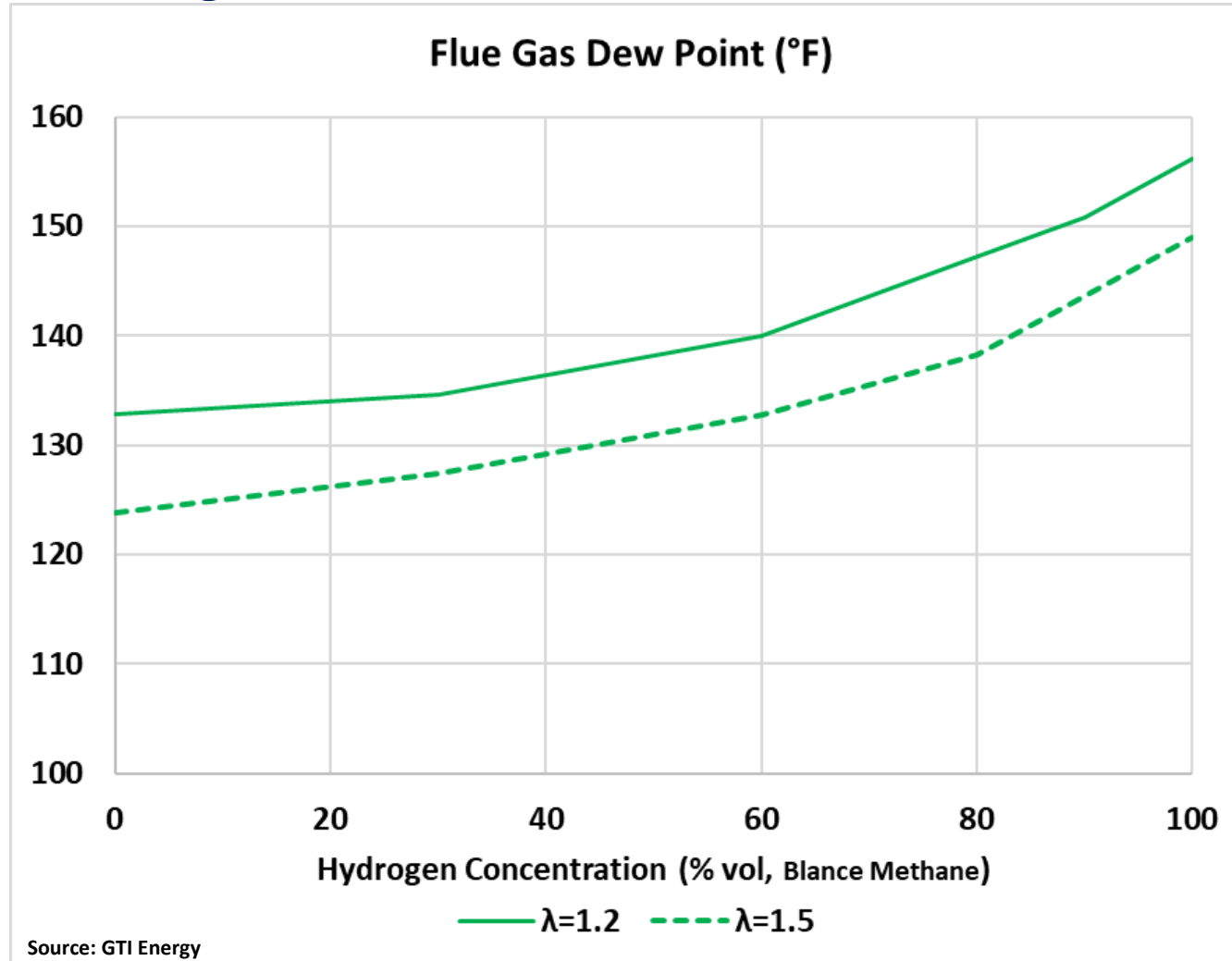


Source: Zhao, McDonell, Samuelsen (2019). Int'l J Hydrogen Energy, Vol. 44(23), 12239



"H₂ 101" - Fundamentals

- Hydrogen generates higher concentration of water in the exhaust stream than does natural gas

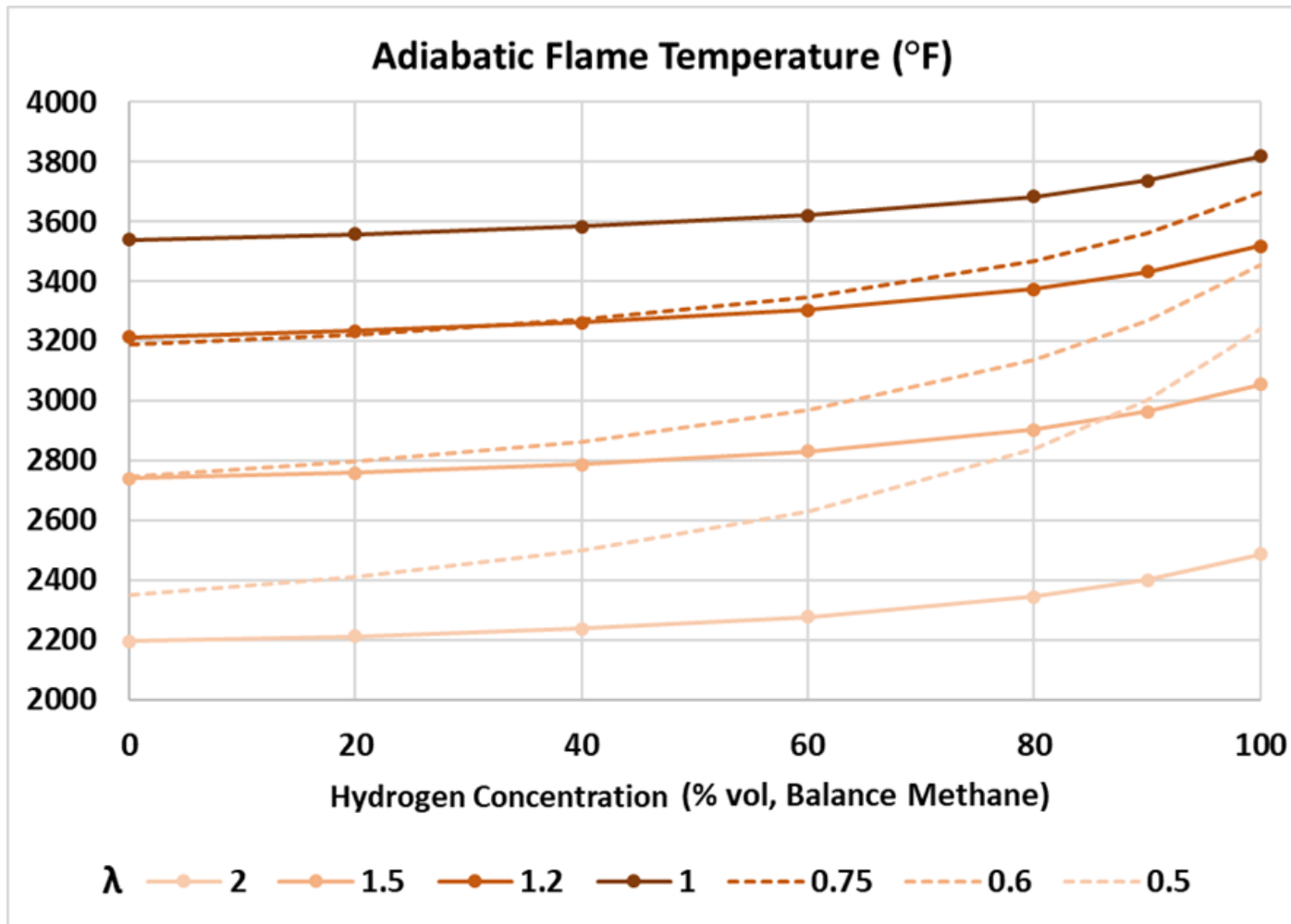


- Consideration for condensing units
- Materials impact/corrosion



“H₂ 101” - Fundamentals

- What about NOx emissions?
 - Yes—Hydrogen has a higher flame temperature for a given fuel air ratio.....but....

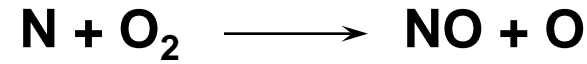
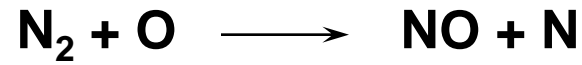


Source: GTI Energy

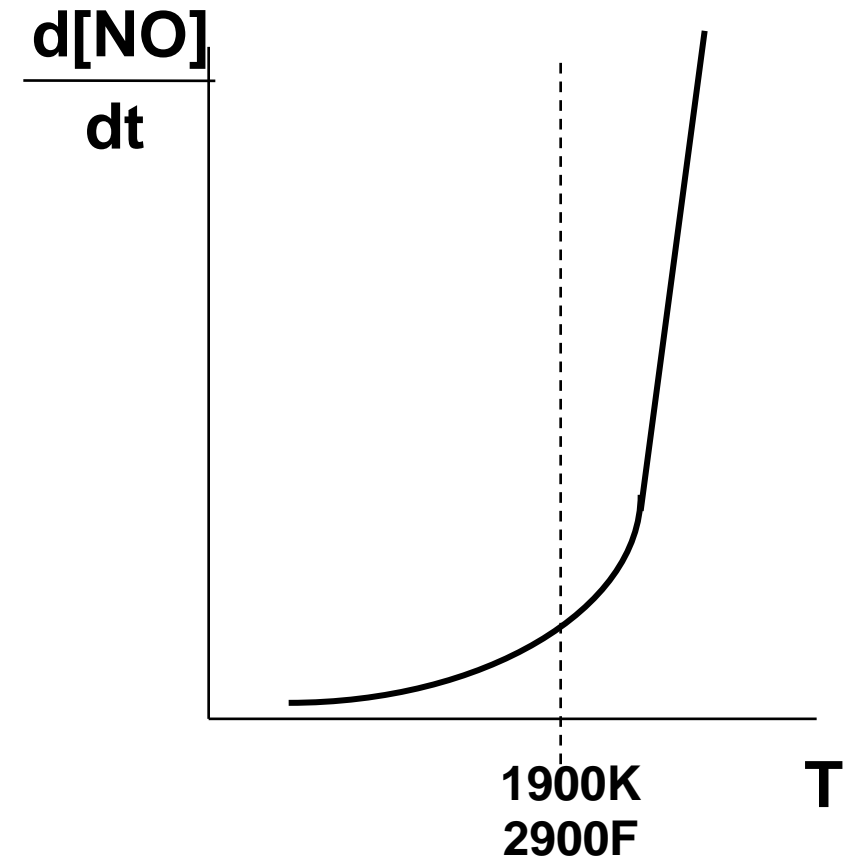
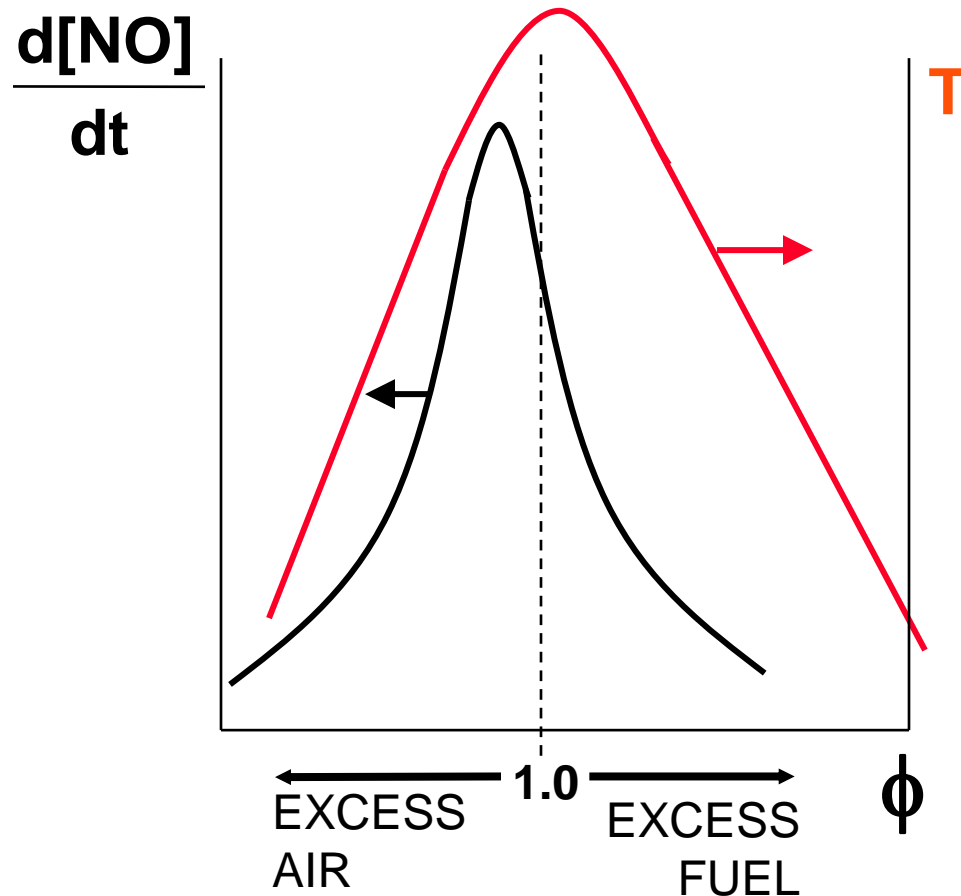


"H₂ 101" - Fundamentals

- NO_x formation background

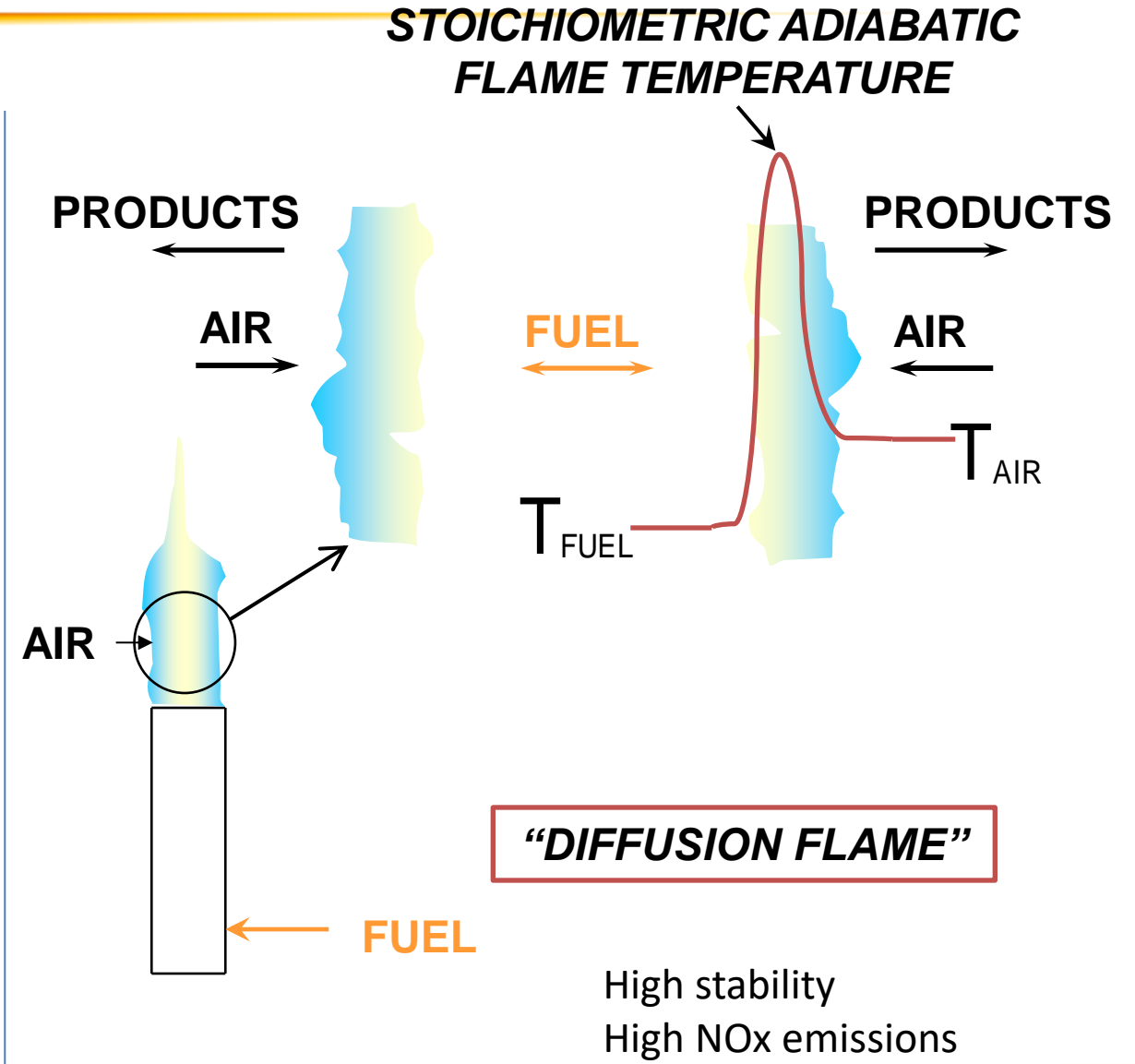
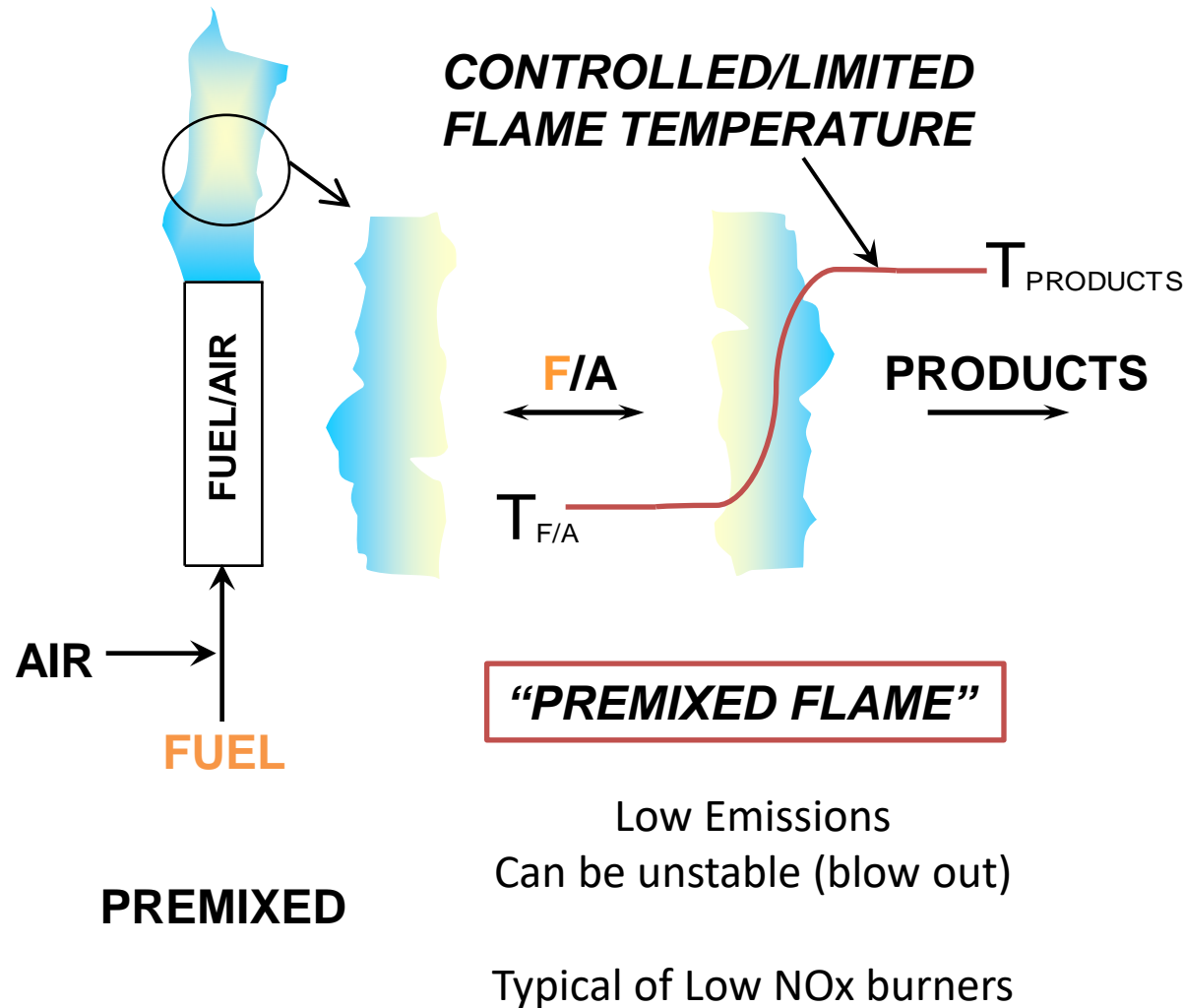


High Combustion Temperatures
→ Enough Energy to Break Triple bonded N₂ molecule



"H₂ 101" - Fundamentals

- Classical Flames



"H₂ 101" - Fundamentals

- Burner Types vary widely depending on applications

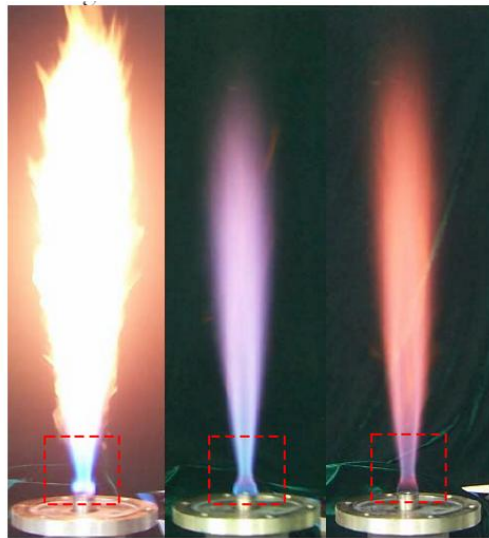
Diffusion

Inject Pure Fuel

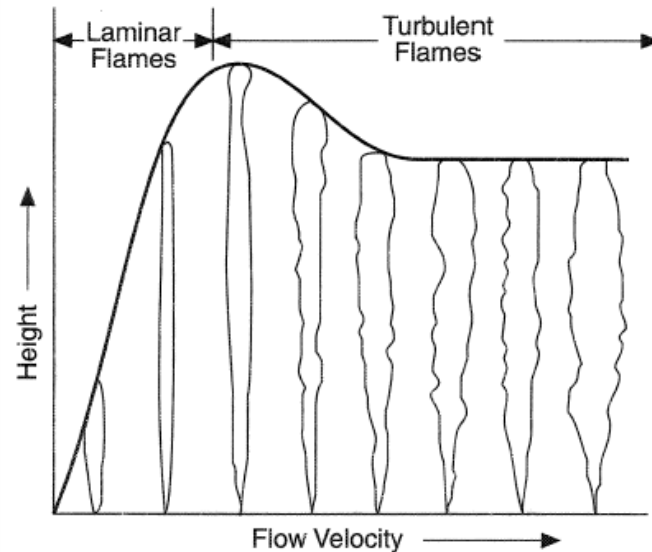
Entrained air mixes and this reacts

Maximum possible flame temps

Turbulence/velocity defines length



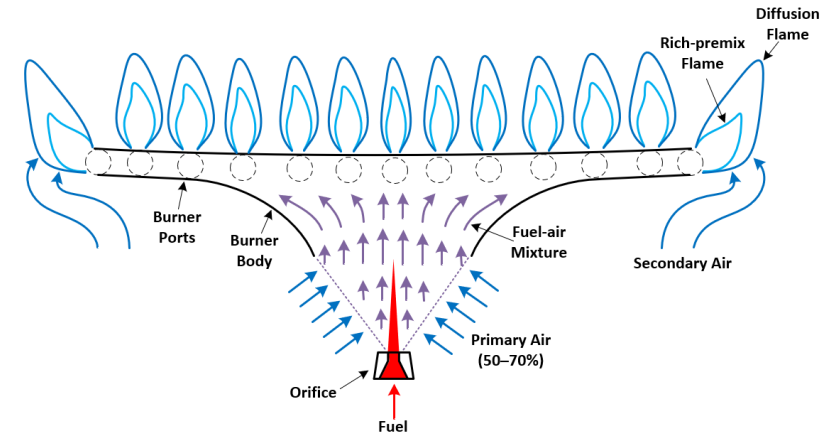
CH₄ 80%H₂+20%CH₄ H₂



Source: Turns

Partially Premixed

Generally, inject fuel rich (too rich to burn*) followed by entrained air to burn out remaining fuel



Source: GTI Energy

*wide flammability of hydrogen is a challenge



“H₂ 101” - Fundamentals

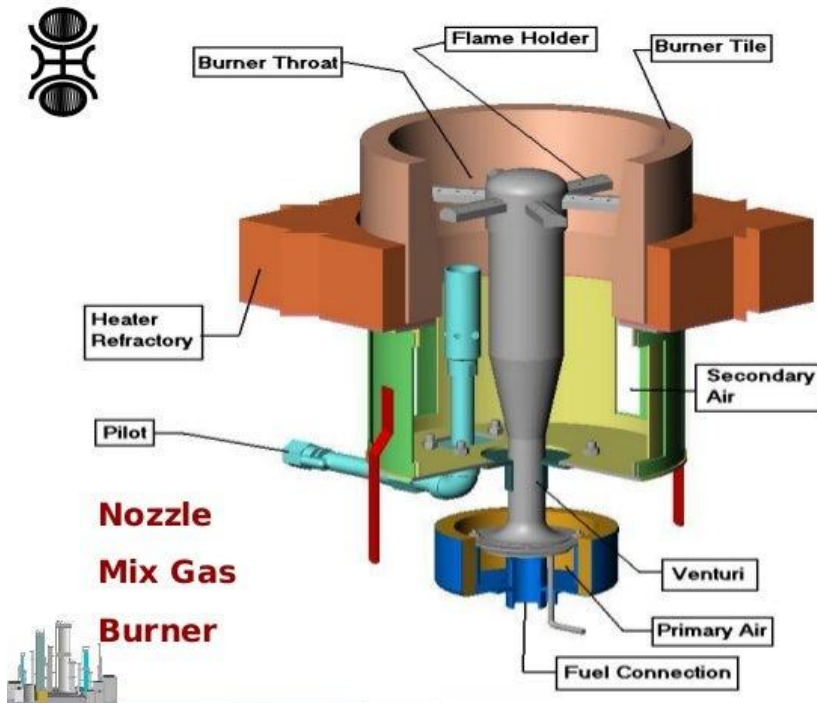
- **Burner Types vary widely depending on applications**

Nozzle Mixed (rapid mix or “LDI”)

Operate overall fuel lean

“rapid mixing” near the point of fuel/air injection

Balance safety and emissions

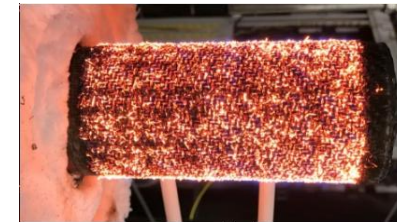
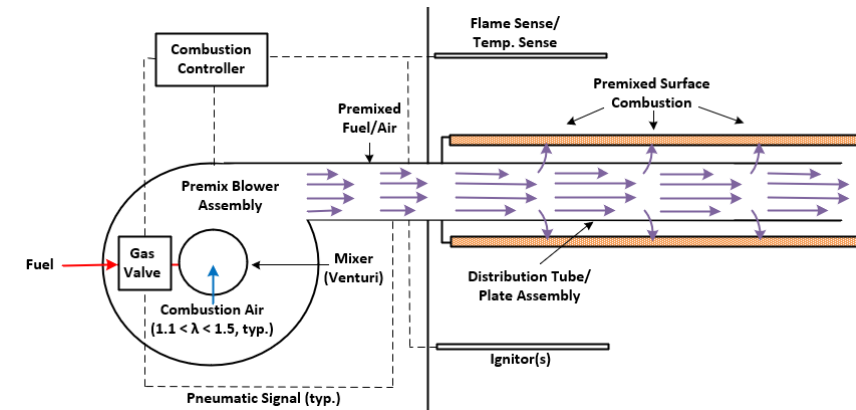


Premixed

Fuel/Air mixed in flammable ratio

“simple” flame temperature control

best NO_x performance

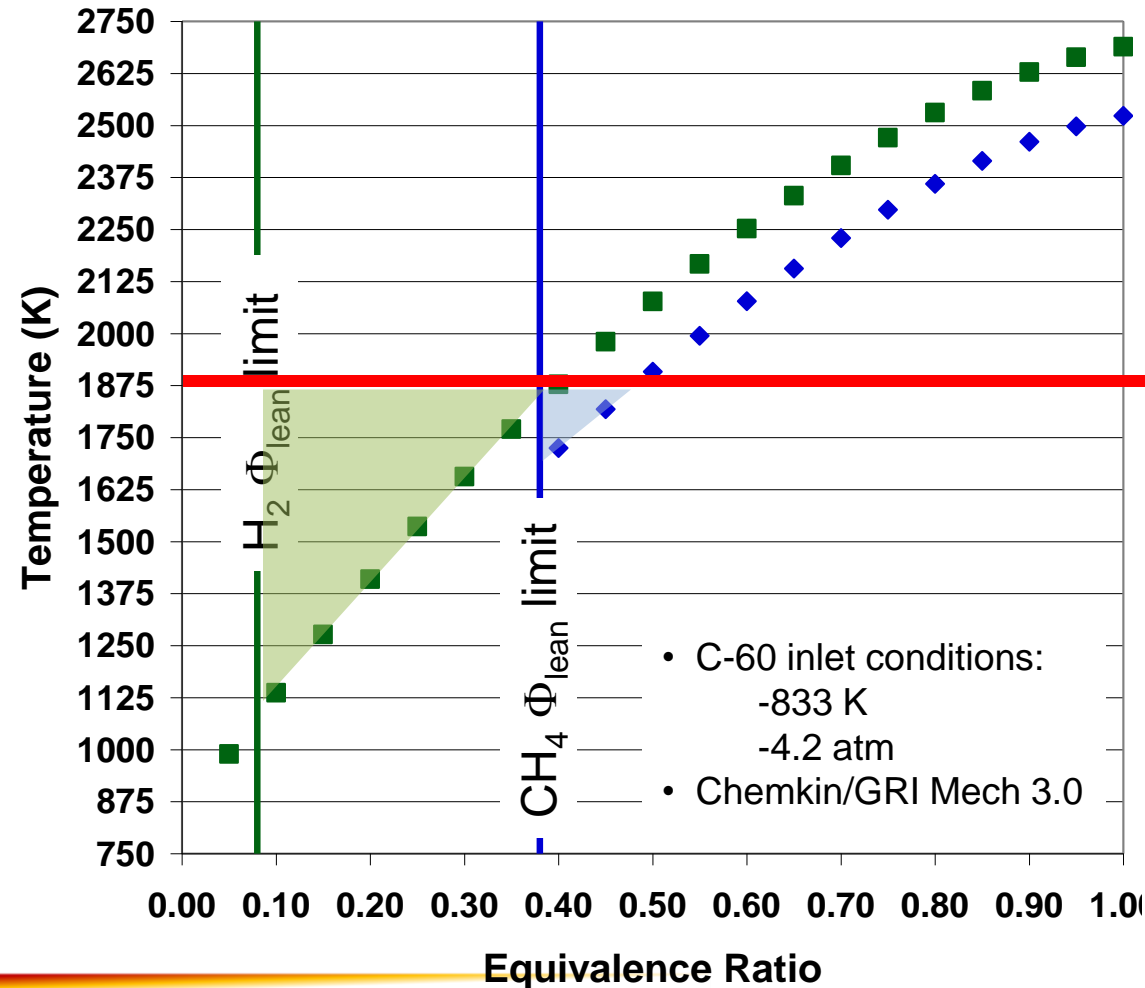


Source: GTI Energy



"H₂ 101" - Fundamentals

- Benefit of wider flammability limits of hydrogen for premixed systems
 - Improved turndown



Hydrogen results in synergy between *improved stability* and *lower emissions*

NOx trigger temperature

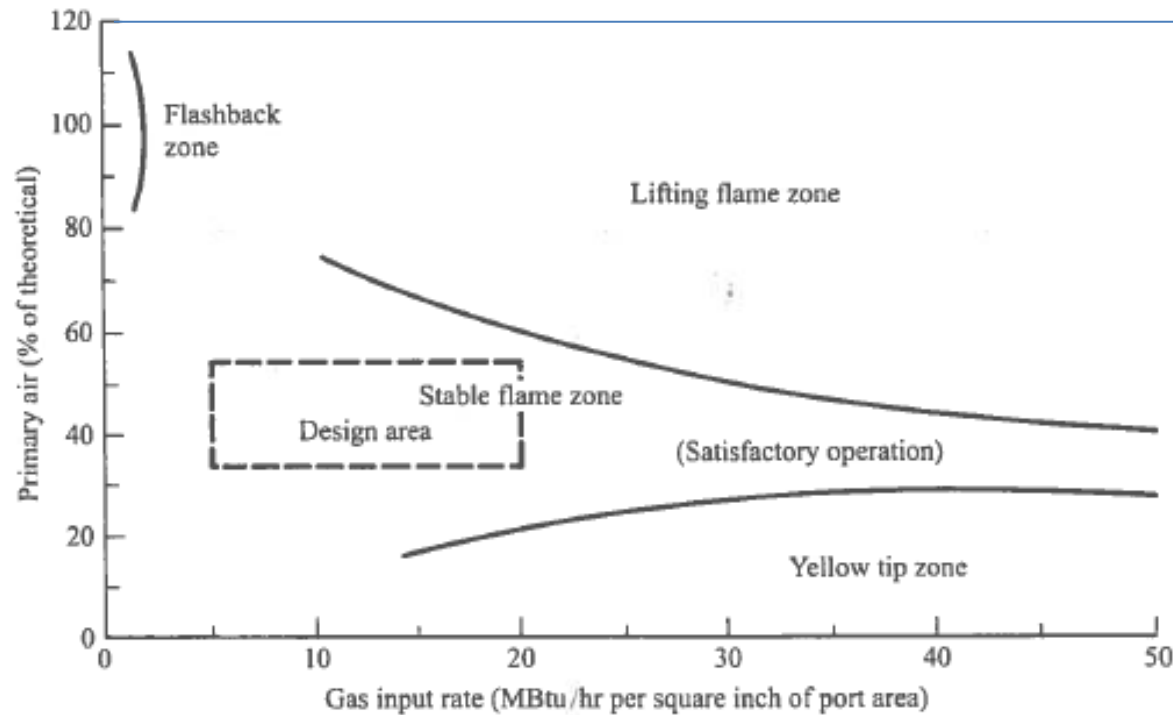


“H₂ 101” - Fundamentals

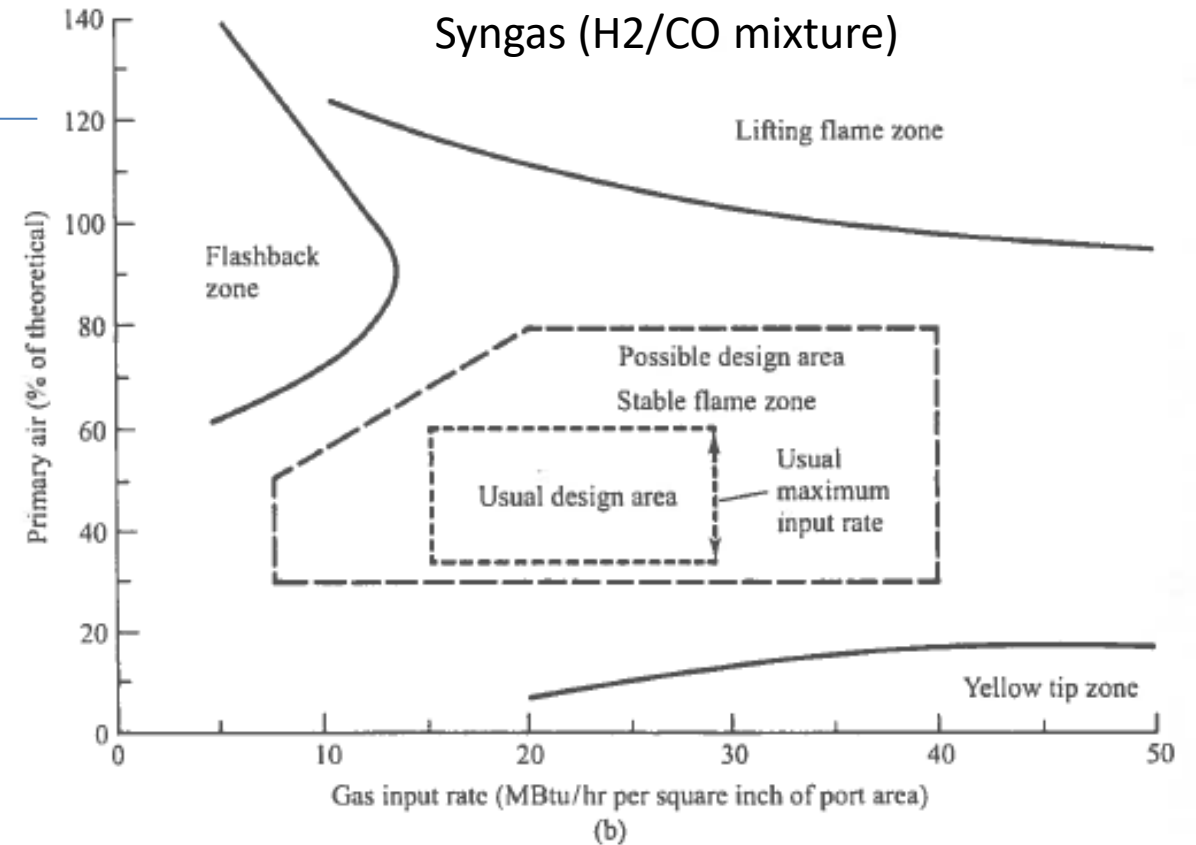
- Operability impacts

- Presence of hydrogen leads to much wider operating limits but with high flashback risk

Natural Gas



Syngas (H₂/CO mixture)

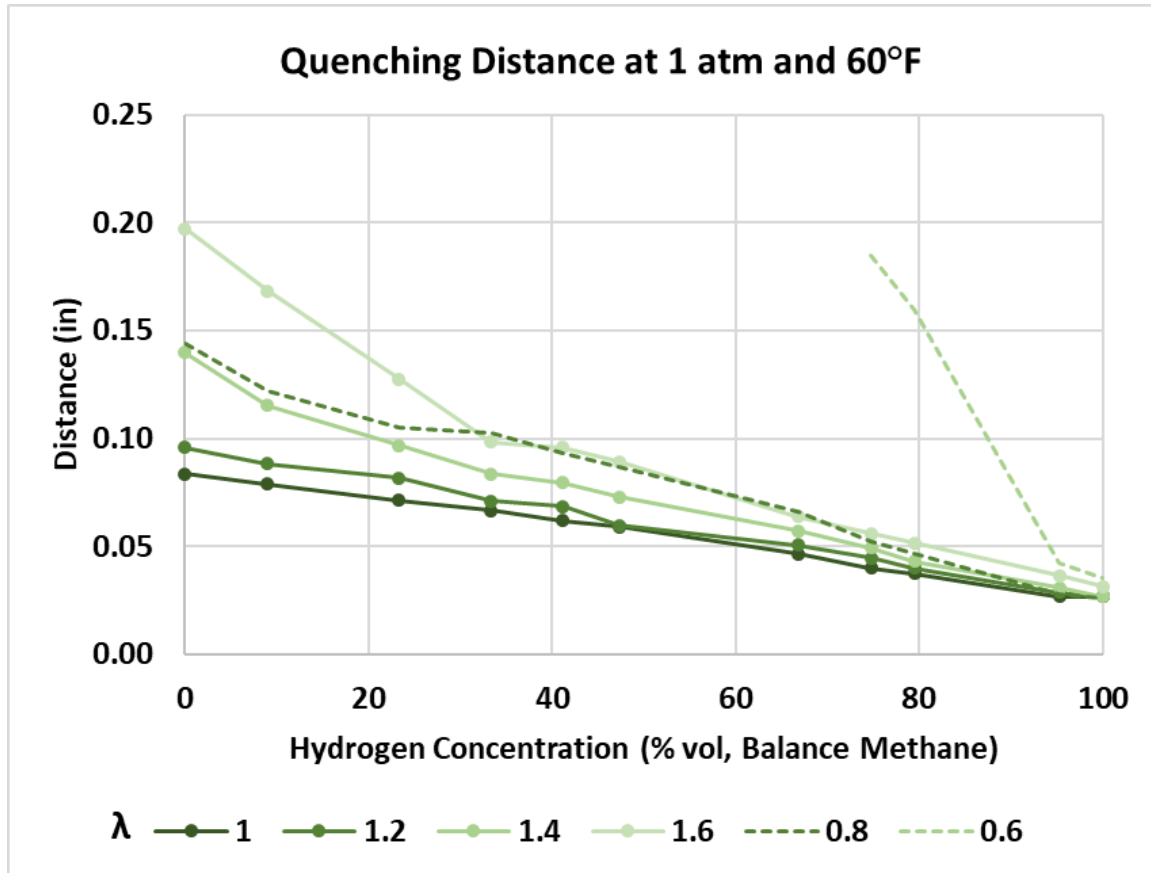


Source: Weber and Vandaveer, Gas Burner Design, 1965

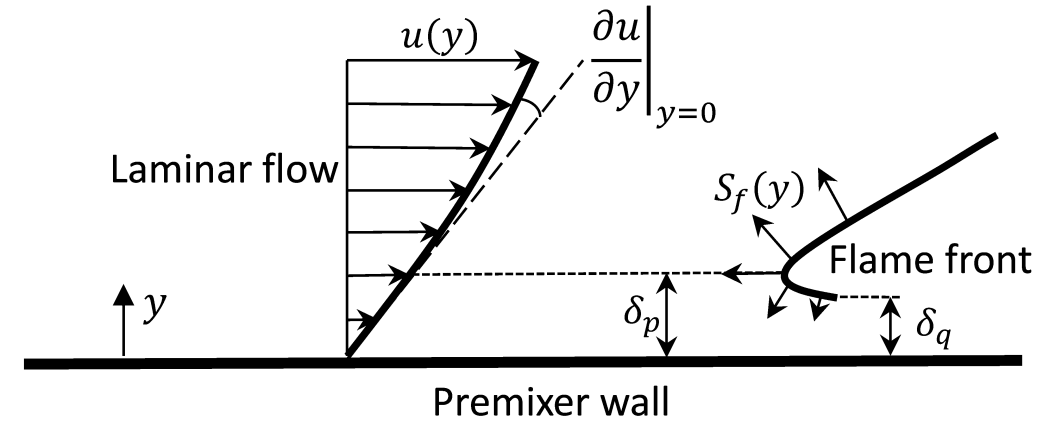


"H₂ 101" - Fundamentals

- Quenching Distance



Source: GTI Energy



Source: Kalantari and McDonell (2017). Prog Energy Combustion Sci., vol 61, 249

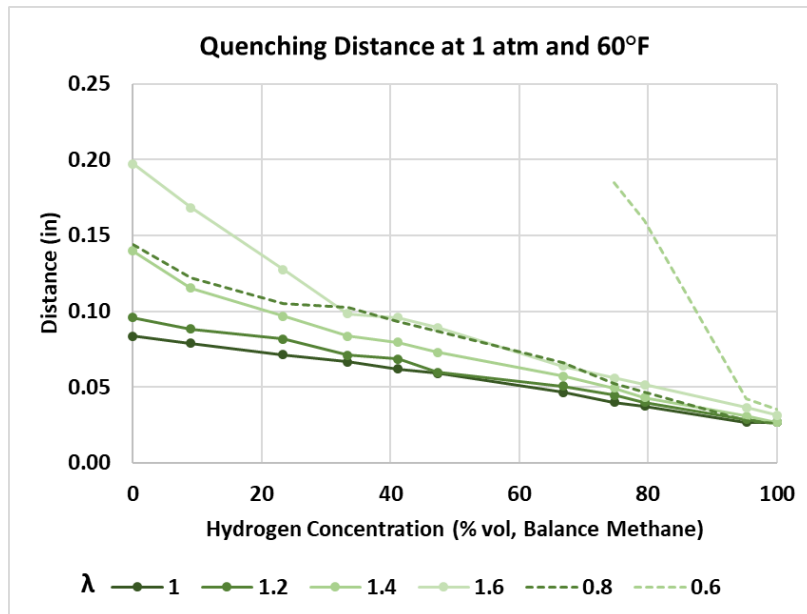
Hydrogen flame can exist much closer to the wall, thus allowing propagation upstream in the boundary layer

Implications for burner holes/mesh, etc

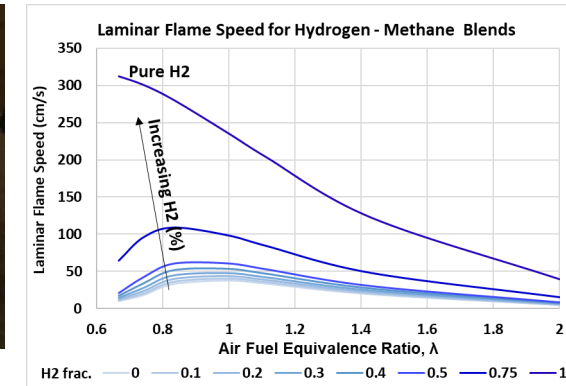
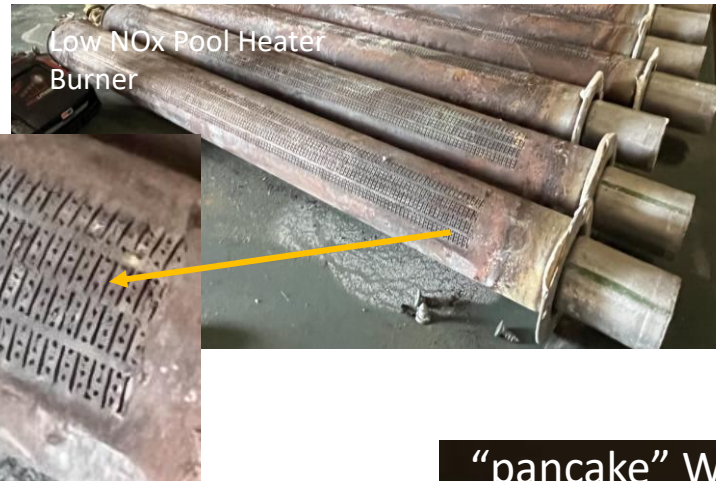


“H₂ 101” - Fundamentals

- For premixed burners, the small quench distance* poses special challenges for hydrogen
 - If flame is not quenched at a port—it will readily propagate (high flame speed)



Source: GTI Energy



Source: GTI Energy



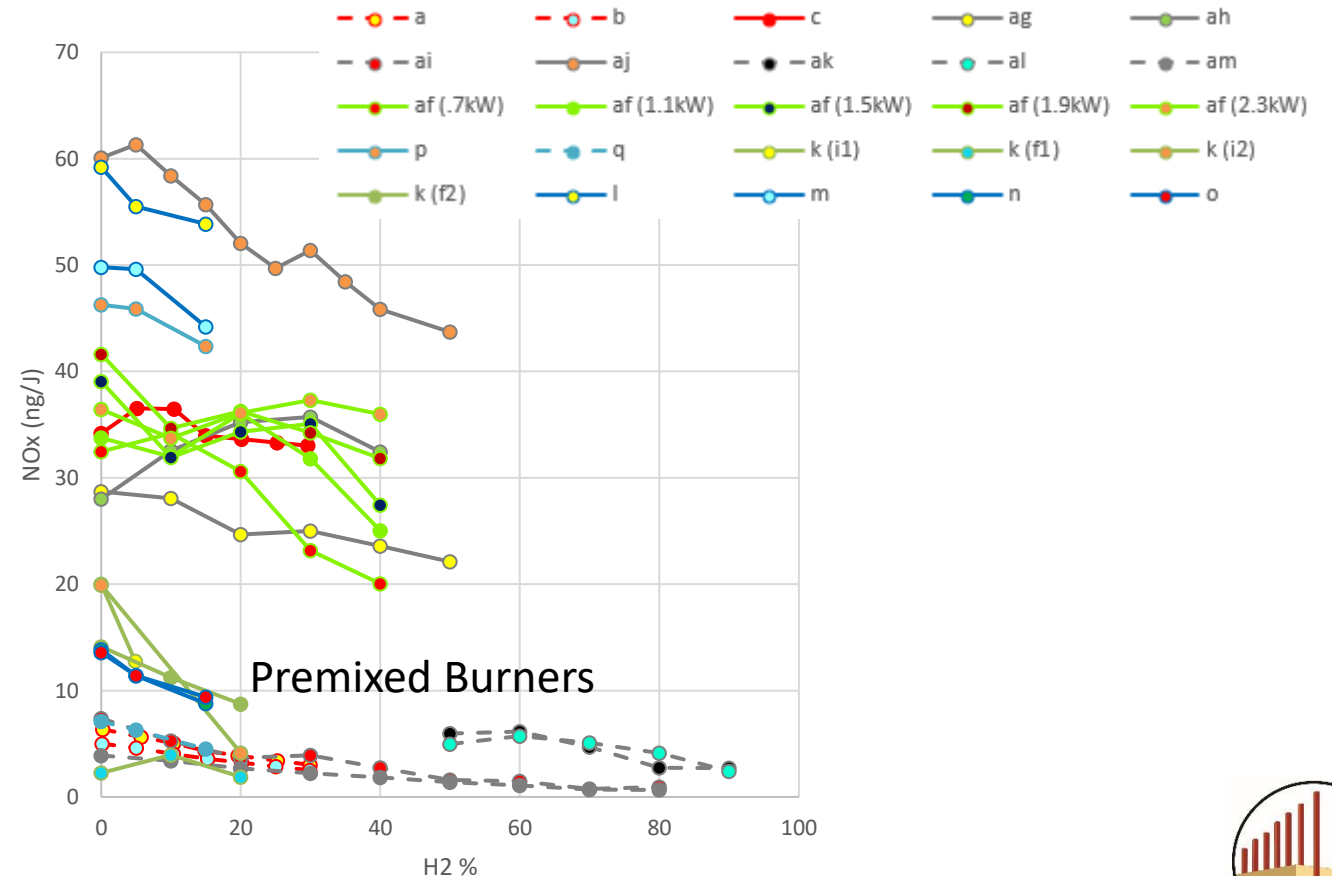
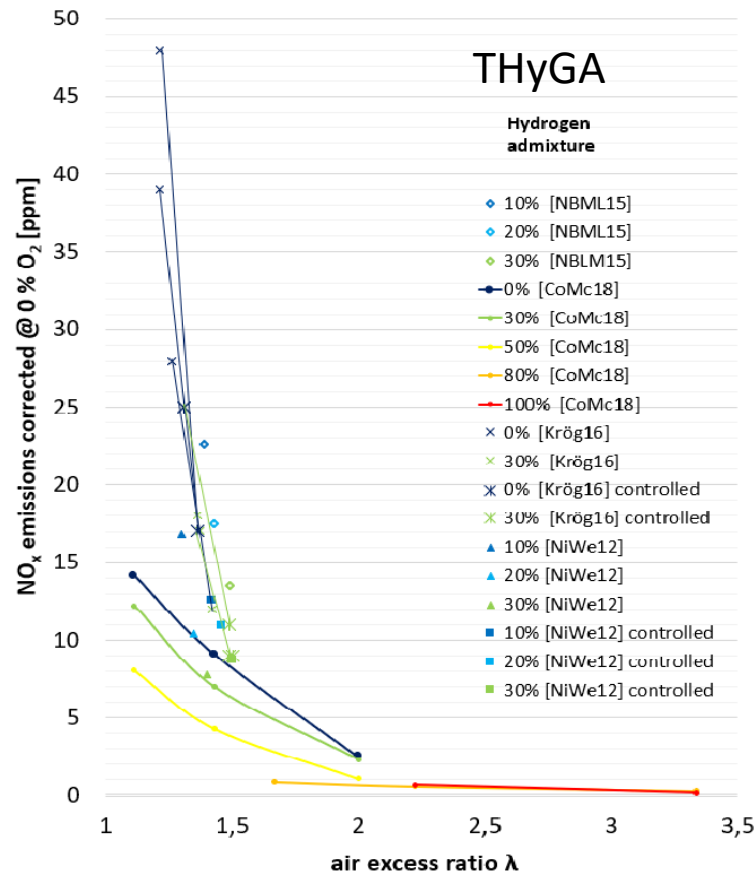
Source: UC Irvine

*Think about burner mesh/ports, etc



"H₂ 101" - Fundamentals

- For premixed burners, NO_x emissions will generally go down
 - Despite higher flame temperature for a given Lambda, because hydrogen needs less air to react, for a given air flow (e.g., entrained volume, forced volume), the operating temperature decreases because lambda inherently increases



Source: Basinger, Hickey and McDonnell (2023). Int. J. Hydr. Energy Vol 48, 19733



“H₂ 101” - Fundamentals

- **Takeaways**

- Hydrogen differs from natural gas
- Flame luminosity
- Flame speed
- Flame temperature
- Wider flammability limits
- Up to some limit of H₂, the fuels behave close enough that modifications not required
 - ✓ (~20-30%)?
 - Limits have been driven by what is being done and is supported by combustion science
 - ✓ Current project is focusing on establishing the upper limit
 - ✓ Transition point from blend to pure hydrogen?



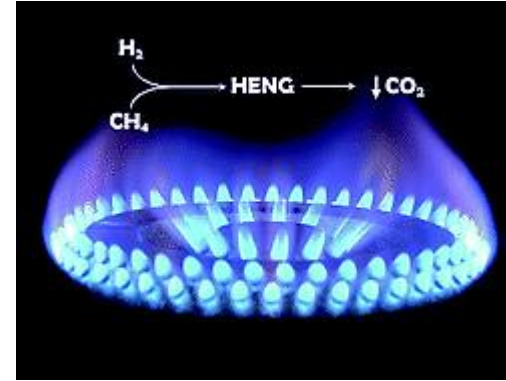
Examining the Effects of Hydrogen in End-use Appliances for Large Commercial Buildings and Industrial Appliances: Test Plan for Long-term Materials Impact



What are concerns from a materials standpoint when burning H_2 in industrial appliances ?

Concerns can be divided to:

- 1- Pre-combustion
- 2- Post-combustion



- Pre-combustion Concerns: there are potential issues with H_2 embrittlement and diffusion into structural materials at higher temperature ($<550^\circ C$) in the combustor.
- Post-combustion concern: There is increased water vapor in the exhaust gas and the potential impact on hot section and downstream components.

Research at UCI will focus on post-combustion concerns from materials standpoint

Post-combustion Environment

H₂ combustion products are different than those from carbon-based fuels. This results in a change in the exhaust gas composition, **such as higher water content.**

	Reactants		Products		Reactants		Products		Reactants		Products	
ϕ	CH ₄ %	H ₂ %	H ₂ O%	O ₂ %	CH ₄ %	H ₂ %	H ₂ O%	O ₂ %	CH ₄ %	H ₂ %	H ₂ O%	O ₂ %
0.95	100	0	18.15	0.96	50	50	21.39	0.94	0	100	33.27	0.88
0.74	100	0	14.42	5.07	50	50	17.06	5	0	100	26.91	4.73
0.50	100	0	9.98	9.98	50	50	11.86	9.88	0	100	19.01	9.51

- Exposing materials to exhaust gases with higher water content, altered composition, and increased temperatures can impact their long-term durability.
- Testing the effects of hydrogen (H₂) on material performance and developing strategies to improve durability is essential before setting safe H₂ addition limits.

Project Objective & Test Methodology

Objective:

- Investigate the effect of blending natural gas with hydrogen (H₂) on the performance of materials used in commercial and industrial combustion systems.

Tested Materials:

- Focus on materials commonly used in industry, such as carbon steel, stainless steel and refractory materials such as alumina/silica.

Testing Conditions:

- Long-term tests ranging from a few hours to 100 hours.
- Temperatures representative of various industries (100 - 1650°C).

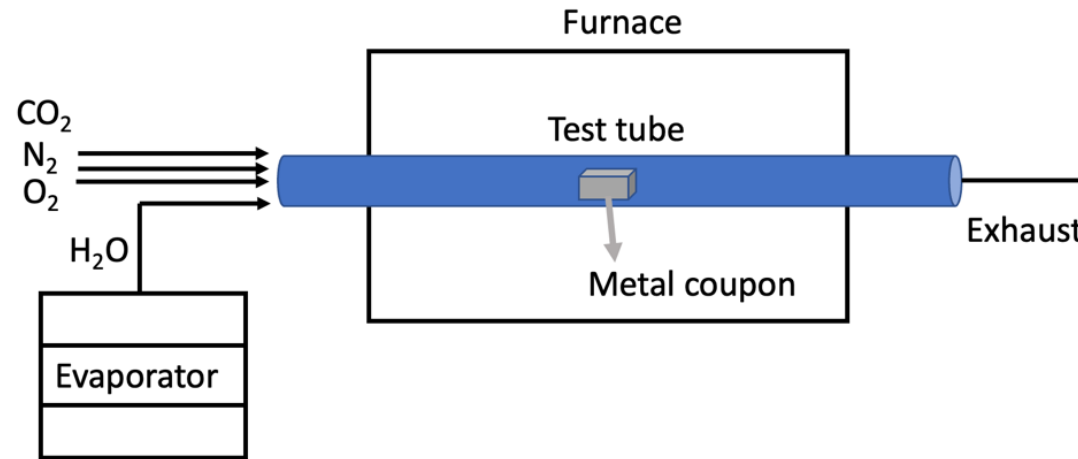
Testing Methodology:

- Coupons of different metals will be exposed to simulated exhaust environments.
- Exhaust composition will vary, focusing on water vapor content to simulate combustion of natural gas and H₂ blends (0 to 100% H₂).

Materials used for different applications

Industry	Burner		Process/Furnace		Exhaust	
	Materials	Temperature	Materials	Temperature	Materials	Temperature
Glass	Cast iron, low or medium carbon steel, SS – 316, Alloys	50-250F (10-121°C)	Refractory – AZS	2000-3000F (1093-1649°C)	Refractory	1200-2000F (649-1093°C)
	Refractory	70-3000 F (21-1648°C)			Carbon Steel	500-1000F (260-538°C)
					Fabric	200-400 F (93-204°C)
Asphalt	Refractory	2500-3000F (1371-1649°C)	Refractory			
	low or medium carbon Steel	50 - 250 F (10-121°C)				
Iron & Steel – High Temp.	Cast iron, low or medium carbon steel, SS – 316, Alloys	50-250 F (10-121°C)	Refractory – Alumina-Silica	2000-2500F (1093-1371°C)	Refractory	750-2000F (399-1093°C)
	Refractory	70-2500 F (21-1371°C)				
Iron & Steel – Mid Temp.	Cast iron, low or medium carbon steel, SS – 316, Alloys	50-250 F (10-121°C)	Refractory	1200-1800F (649-982°C)	Carbon Steel	500-1000F (260-538°C)
	Refractory	70-500 F (21-1371°C)	Fiberboard	1200-1800F (649-982°C)	Alloy	500-1500F (260-538°C)
			Alloy	1200-2000F (649-1093°C)		
			Ceramic – Si-SiC	1500-2300F (816-1260C°)		
Food and Beverage	low or medium carbon Steel	70-1000F (21-538°C)	Carbon Steel	400-500F (204-260°C)	Carbon Steel	400-500F (204-260°C)
	Ceramics					
	SS – 304, 316					
Chemicals – Catalytic Cracking	Cast iron, low or medium carbon steel, SS – 316, Alloys -	50-250F (10-121°C)		900F (482°C)	Carbon Steel	400-500F (204-260°C)
	Refractory	70-2500F (21-1371°C)			Refractory	750-2000F (399-1093°C)
Refining		2000F (1093°C)	25Cr-35Ni	1550F (843°C)		1000-1350F (538-732°C)
Paper and Pulp – Boilers mainly	Boilers	2500F (1371°C)	Carbon steel, cast components	120-350F (49-177°C)	High grade carbon steel	200F (93°C)
Cement	Carbon Steel	50-250F (10-121°C)	Refractory	2000-3000F (1093-1649°C)	Refractory	500-2000F (260-1093°C)
	Refractory	2000-3000F (1093-1649°C)				

Experimental Set-up



Experimental Set-up:

- Metal coupons will be heated in a test tube using a Thermcraft split furnace, with temperatures up to 1650°C to simulate high-temperature burner conditions.
- Exhaust gas will be composed of controlled mixtures of CO₂, O₂, N₂, and water vapor.
- Water vapor content will be adjusted to simulate exhausts of natural gas/H₂ blends.

Safety Features:

- Gas sensors in the walk-in hood detect leaks and activate a control panel that Shuts down gas supply from tanks, increases room ventilation, triggers visual and audible alarms.

Materials Characterization

Characterization Facility:

- Materials characterization will be conducted at UCI's Irvine Materials Research Institute (IMRI), equipped with advanced instrumentation for materials analysis.

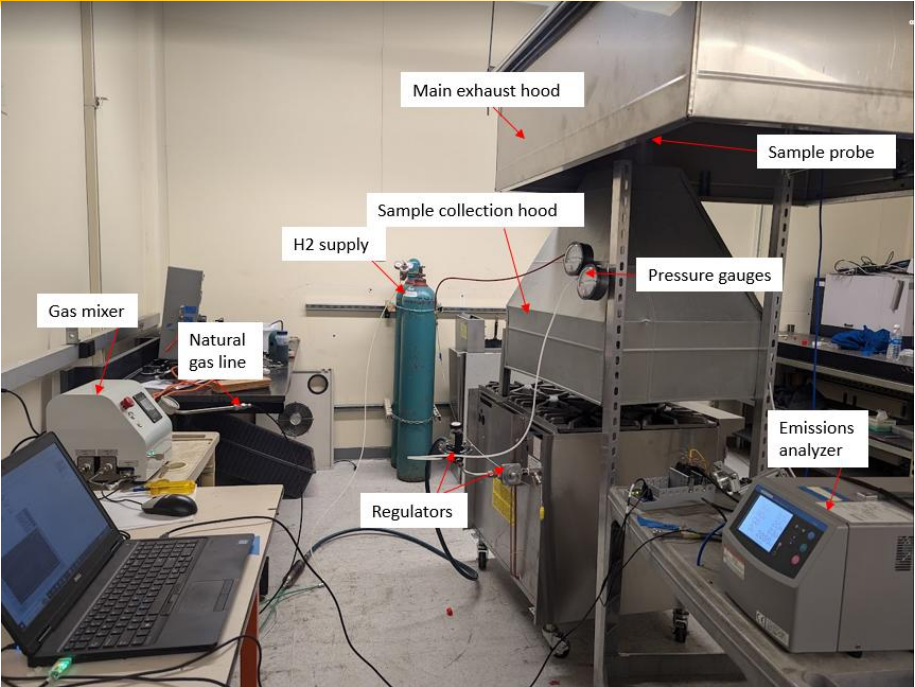
Characterization Techniques:

- **XRD:** Analyze changes in the crystal structure.
- **SEM:** Observe material structures and detect cracks, supported by elemental mapping with EDS.
- **FIB:** Prepare cross-sections of samples.
- **STEM:** Image cross-sections with EDS mapping for detailed elemental analysis.

Objective:

- These techniques will assess the impact of exhaust gases from natural gas/H₂ blends on material properties.

Combustion Performance: Devices Under Study

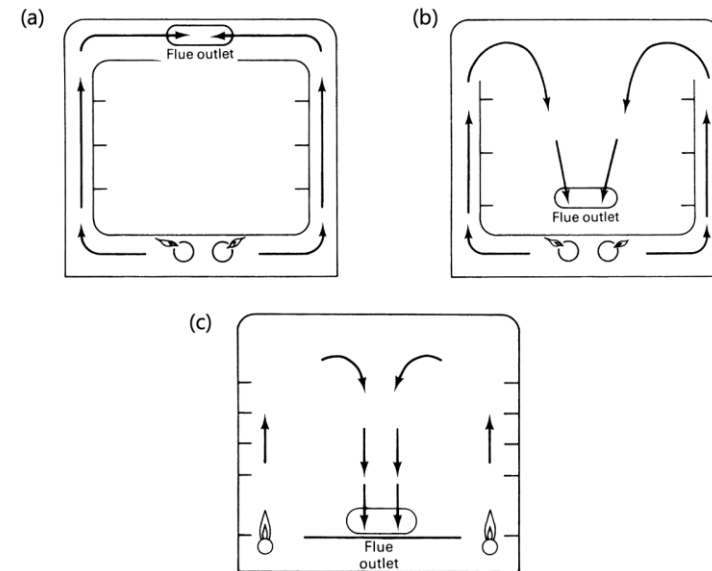
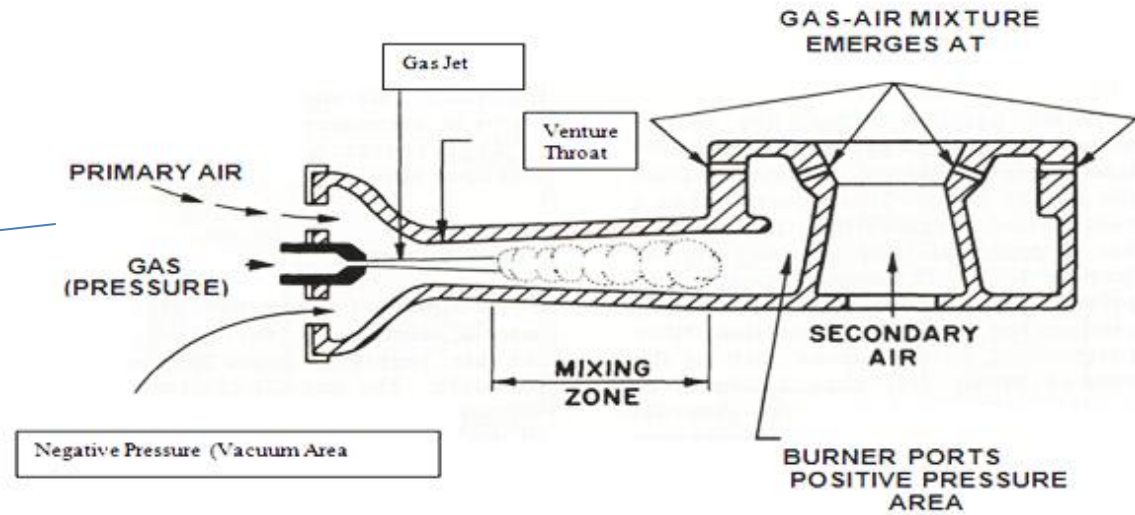


Performance Testing Example: Cooking Ranges

- Stovetop/Range



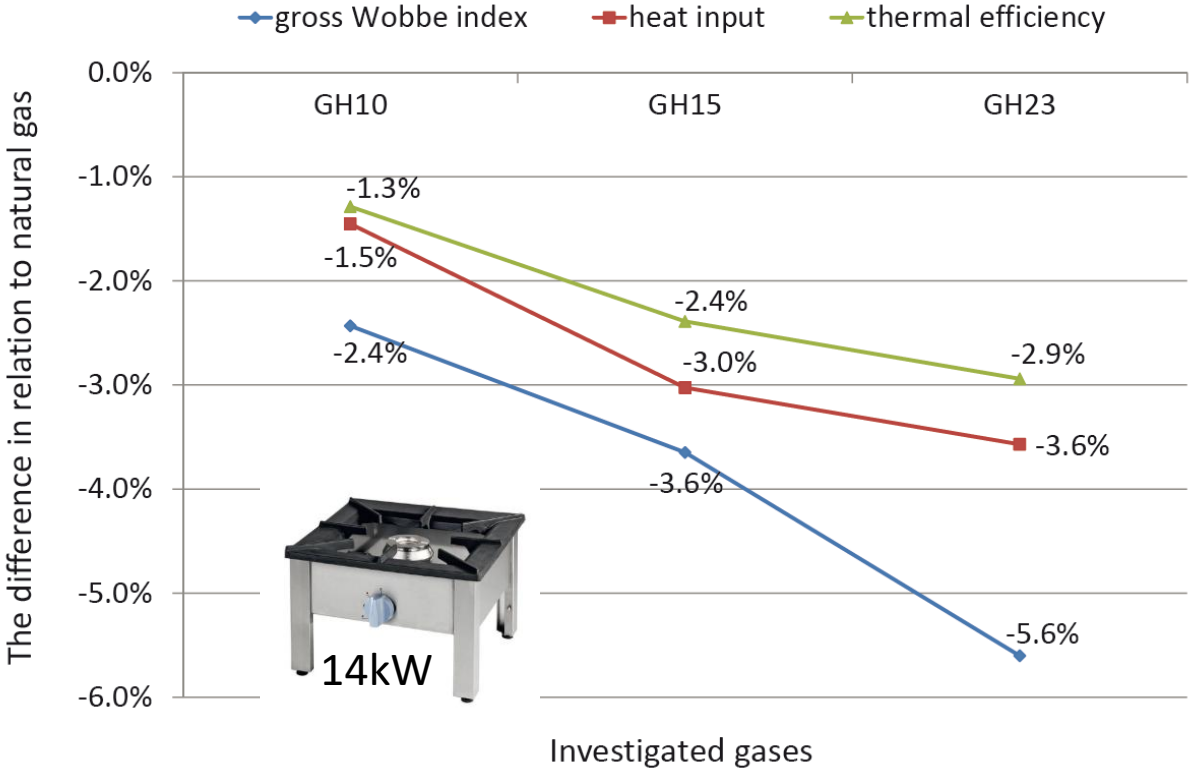
Cooking Ranges/Ovens



Ranges

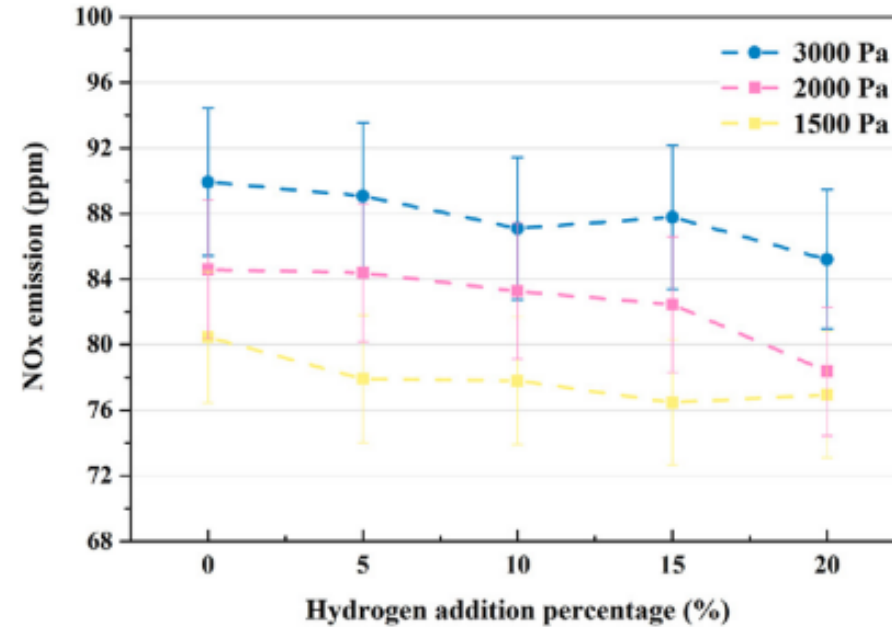
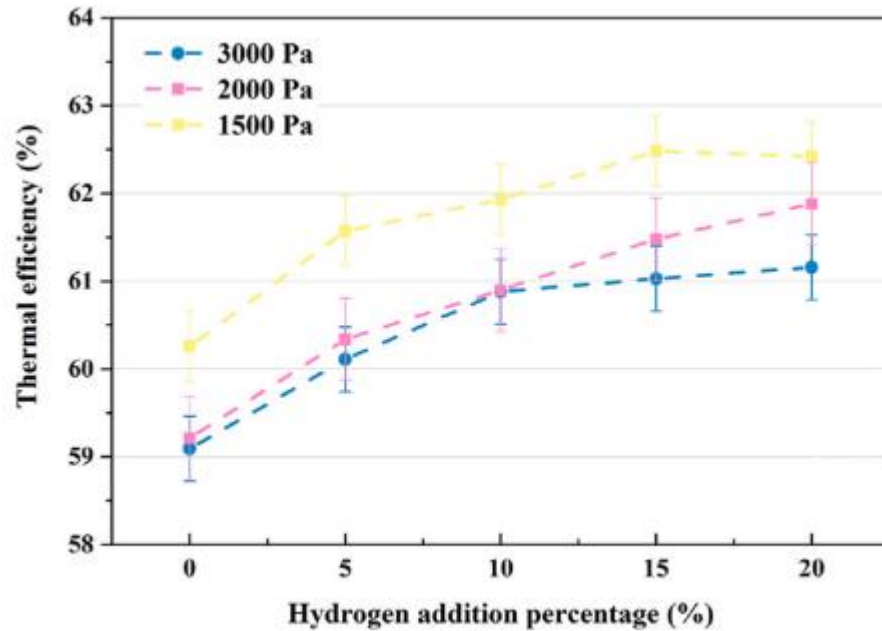
- Previous Work: Gas Ranges
 - Commercial cooking stock pot range (Wojowicz, 2019)
 - Up to 23% hydrogen added

Parameter		Unit	Gas symbols and parameter's value			
			2E*	GH10	GH15	GH23
Gas components	hydrogen		–	10.01	15.01	23.00
	methane	%	97.34	87.59	82.73	74.95
	nitrogen	%	1.15	1.04	0.98	0.89
	others	%	1.51	1.36	1.28	1.16
Gross calorific value H_s		MJ/m ³	37.90	35.32	34.03	31.96
Net calorific value H_i		MJ/m ³	34.17	31.78	30.58	28.66
Gross Wobbe index W_s		MJ/m ³	50.16	48.94	48.33	47.35
Net Wobbe index W_i		MJ/m ³	45.26	44.04	43.44	42.47
Density ρ		kg/m ³	0.700	0.638	0.607	0.558
Relative density d		–	0.571	0.521	0.496	0.456



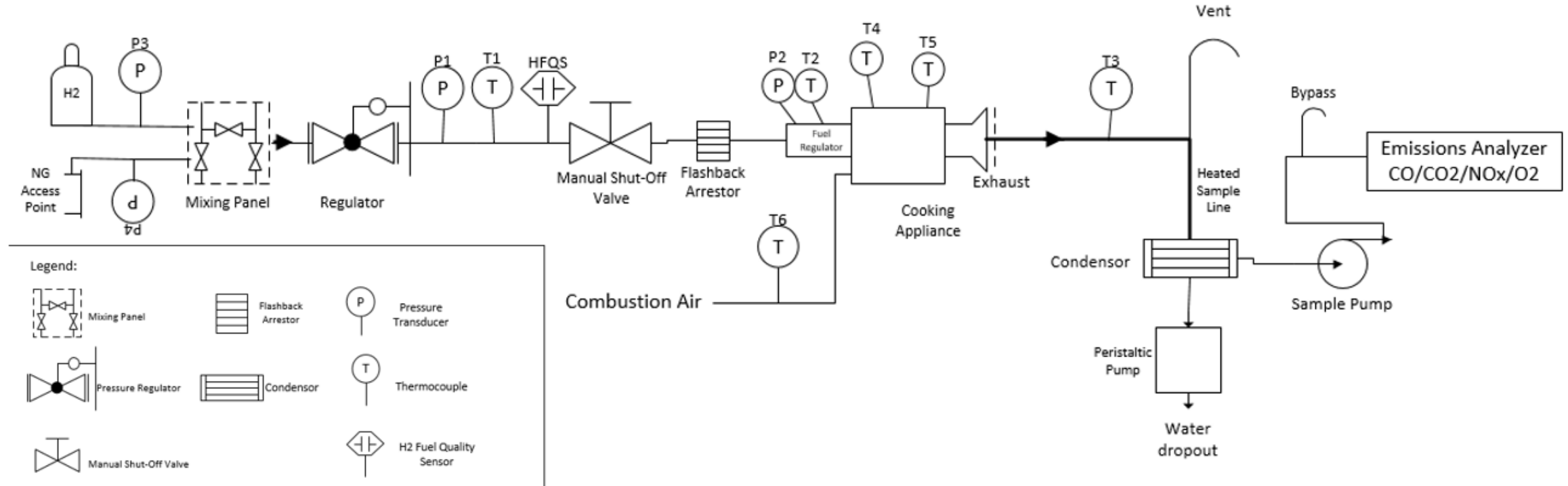
Ranges

- **Previous Work: Sun et al., 2022—up to 20% hydrogen added by volume**
 - Efficiency increases with hydrogen addition
 - NOx emission generally decrease



Performance Results: Current Effort

- Test Setup



Performance Results: Current Effort

- **Test Plan**

- **Steady Combustion Test**
- **Ignition Test**
- **Dynamic Blending (“slug” test)**
- **ASTM F1521-22 Efficiency Test**

- **Base case with matched volume flow (dictated by maintaining gas pressure specs for fuel inlet)**
 - ✓ **Necessarily results in reduced heat input to device (Wobbe Index)**
- ***If feasible*, matched heat input (requires override of controls, higher pressures, etc.)**
 - ✓ **This can be done via adjustments in metering and controls in the field if needed**
 - ✓ **If limited H₂ addition (~20%) this can be done**

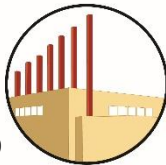
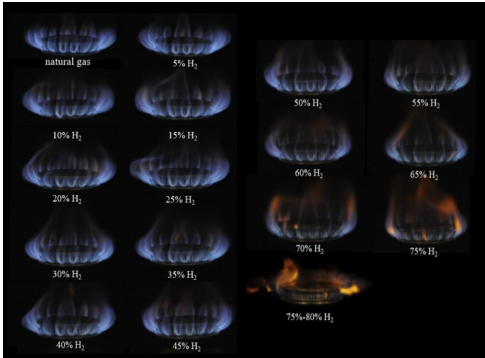


Commercial Ranges

- Ignition and Dynamic Blending Results
 - Commercial devices are tolerant to H2 addition

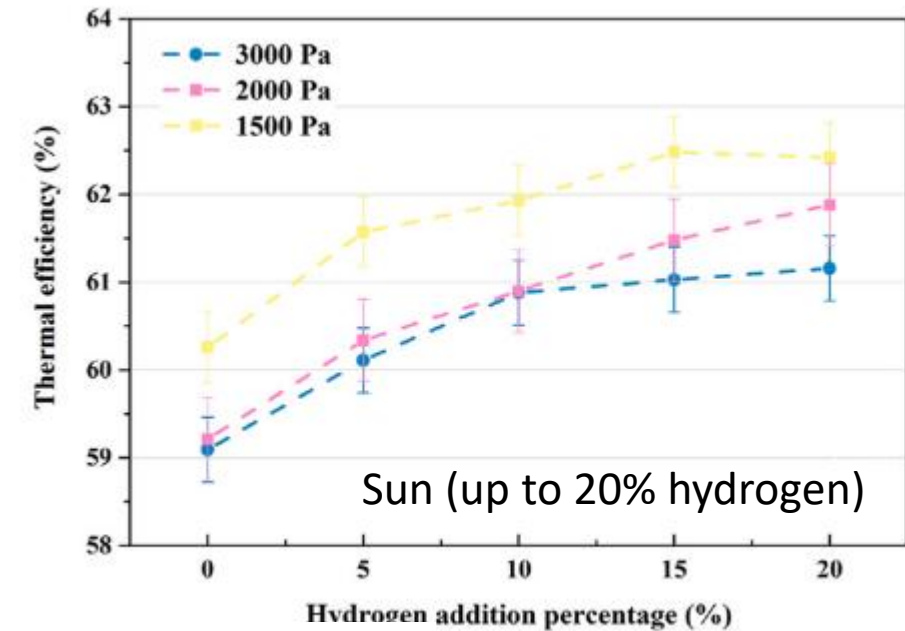
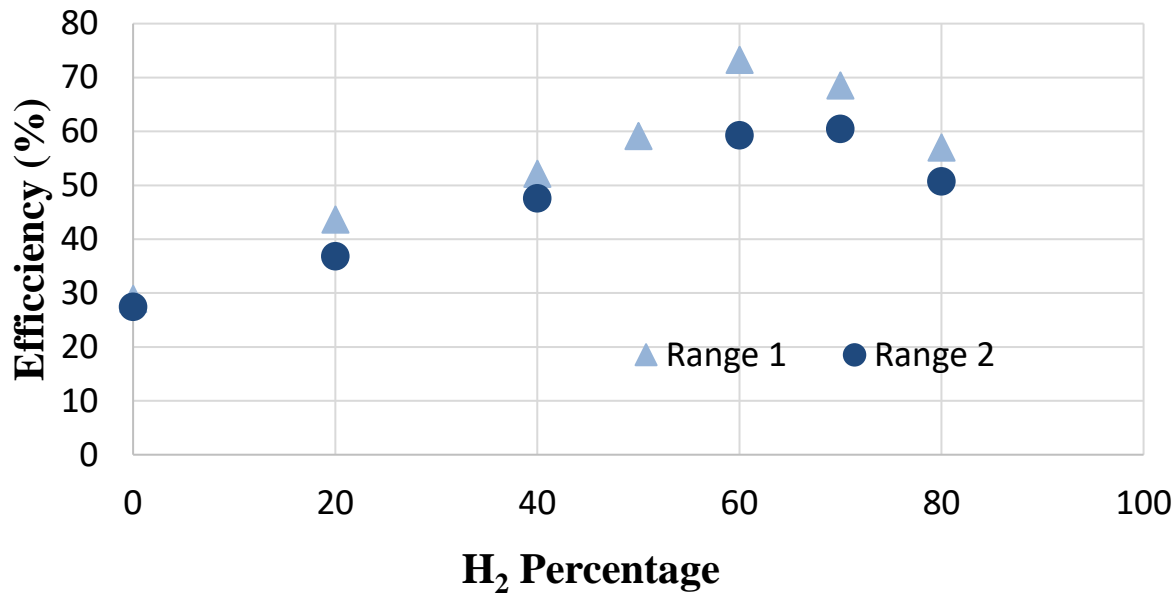
Appliance	Heat Setting	Flashback At (%H ₂)			Dynamic Blending Test (%H ₂)
	(High/low)	Steady-state	Hot Cycle	Cold Cycle	
Range 1	low	95	NA	NA	90
	high	90	90	70	80
Range 2	low	90	NA	NA	80
	high	95	95	95	90

- Residential Systems exhibit issues at 45-50%
 - ✓ more tolerance if starting on NG and then increasing H2 content



Commercial Ranges

- Efficiency
 - ASTM F1521-22 Efficiency Test



$$\eta_{cook} = \frac{E_{water} + E_{pot}}{E_{input}} \times 100\%$$

where:

$$E_{water} + E_{pot} = ((W_{water} \times C_{p_{water}}) + (W_{pot} \times C_{p_{pot}})) \times (T_2 - T_1)$$

W_{water} = weight of water in the sauce pot, that is specified as 20 lb (9091 g) of water,

$C_{p_{water}}$ = specific heat of water = 1.0 Btu/lb·°F (418.7 J/kg·°K),

W_{pot} = weight of cooking container, as specified in 6.3,

$C_{p_{pot}}$ = specific heat of cooking container, specified as either: aluminum = 0.22 Btu/lb·°F, or steel = 0.11 Btu/lb·°F,

T_2 = ending temperature of the water, that is specified as 200°F (93°C),

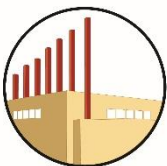
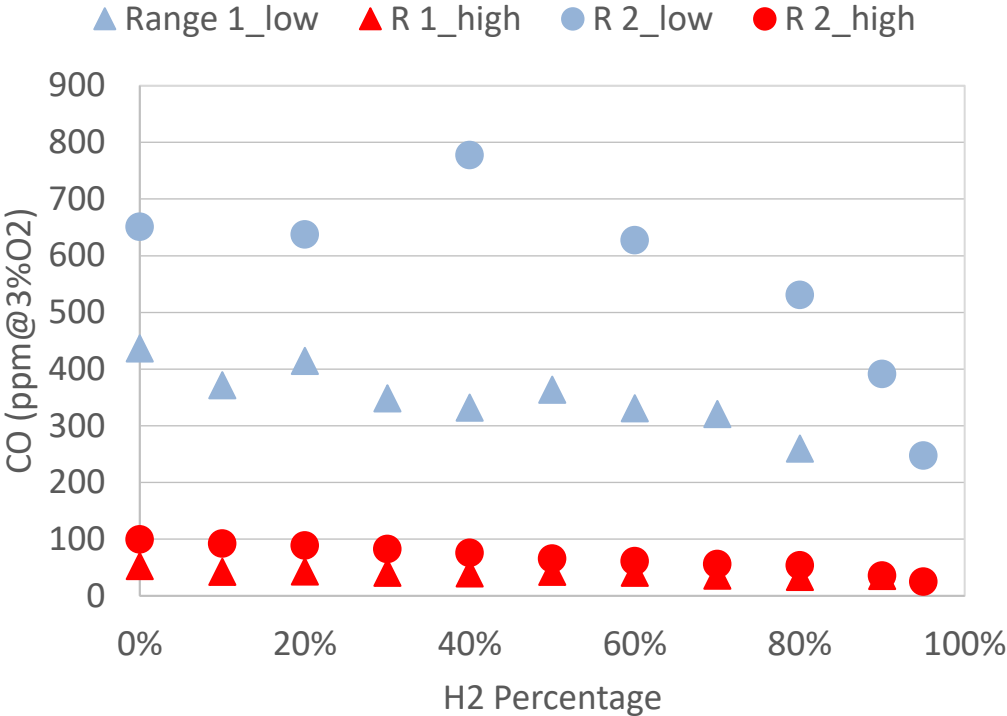
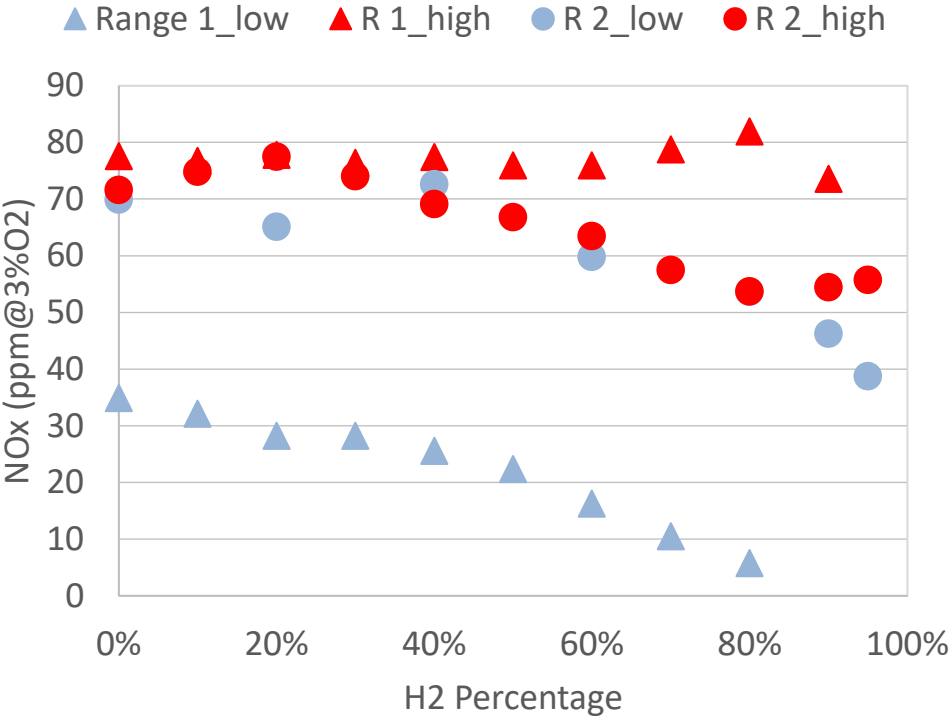
T_1 = beginning temperature of the water, that is specified as 70 ± 2°F (21 ± 1°C), and

E_{input} = energy consumed by the cooking unit during the test, Btu, including any electric energy consumed by a gas range top.



Commercial Ranges

- Emissions



General Tolerance of Commercial Devices

- Ranges, Ovens, Griddles, Fryer
 - Operability

Appliance	Heat Setting	Flashback At (%H₂)			Slug Test
	(High/low)	Steady-state	Hot Cycle	Cold Cycle	
Range 1	low	95			90
	high	90	90	70	80
Oven 1	low	95	90	90	90
	high	90	90	90	80, 85
Range 2	low	90			80
	high	95	95	95	90
Oven 2	low	95	90	70	90
	high	90	90	70	80
Griddle 1	low	100			95
	high	95	95	90	90
Griddle 2	low	80	80		70
	high	70	70	70	60
Fryer 1	Low	100	100	100	90

- In comparison to residential appliances the devices studied here have more tolerance to hydrogen addition



Example of Flashback

- Oven Burner “failure” mechanism

Example of Flashback for Oven Burner

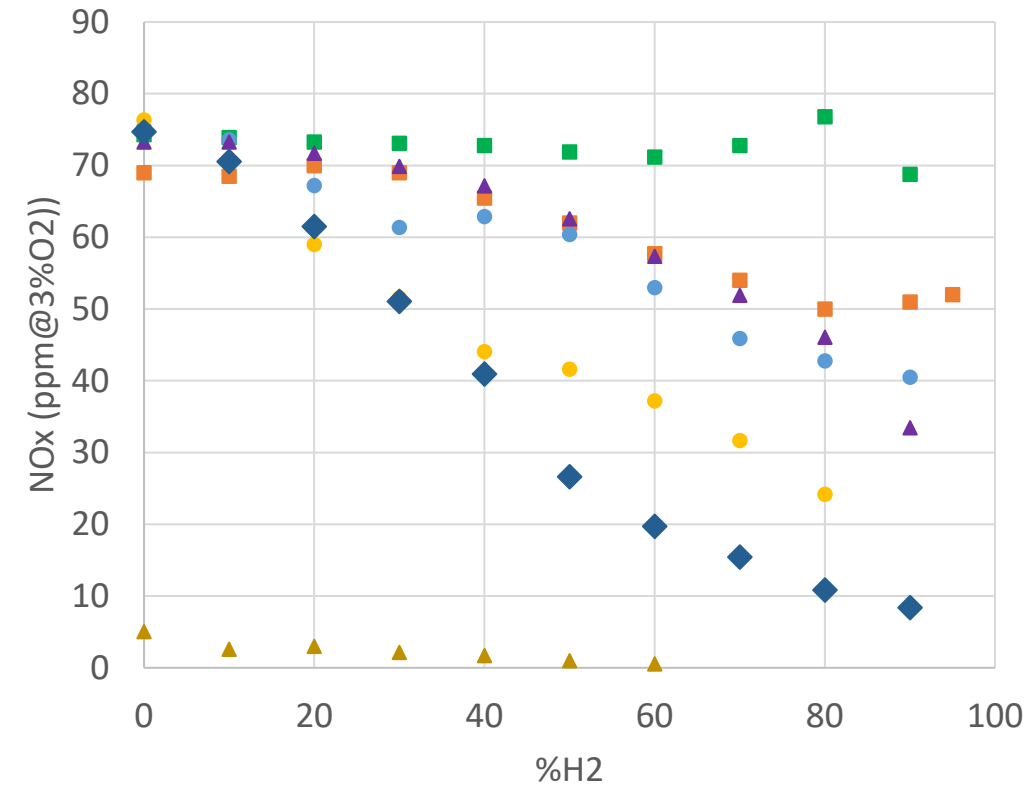
90-95% Hydrogen



Summary of Emissions Performance for Cooking Appliances

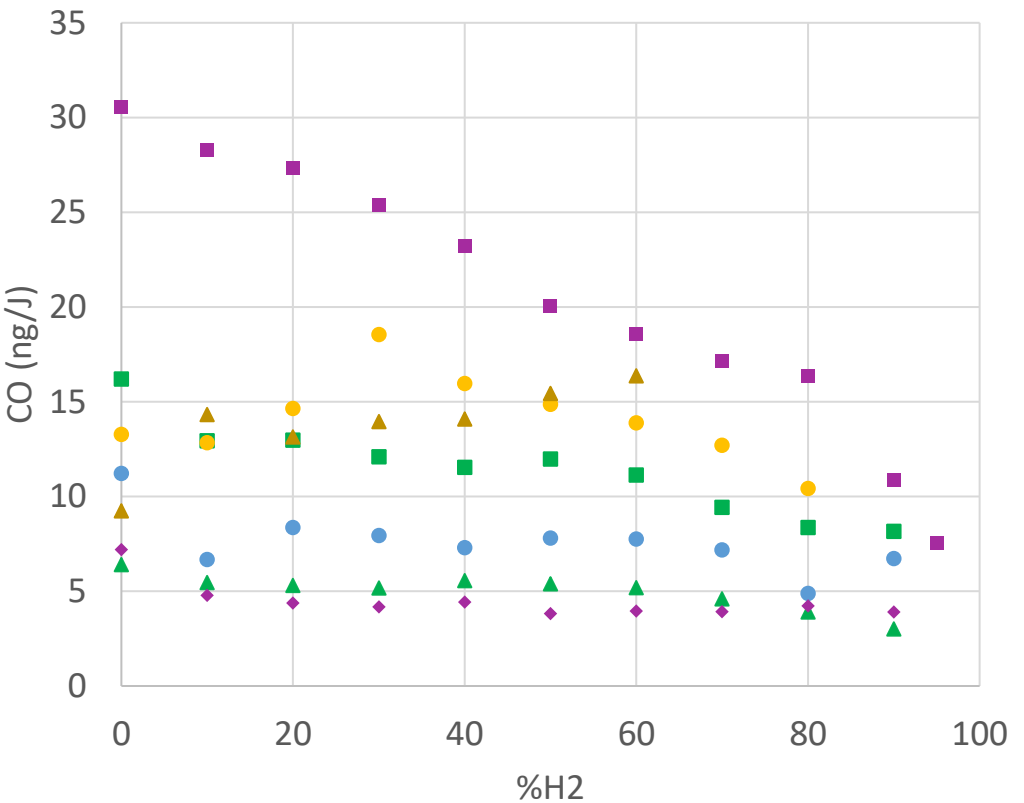
- Range, Oven, Griddle, Fryer

NOx ppm (high heat condition)



Range 1 Range 2 Oven 1 Oven 2
Griddle 1 Griddle 2 Fryer 1

CO ppm (high heat condition)



Range 1 Range 2 Oven 1 Oven 2
Griddle 1 Griddle 2 Fryer 1



More to come.....



Planning to test Units
in UCI Combustion Lab



Water Heaters
Dryer Units

Cooking Units (Classroom)

Building Heater
Water Heaters

Pool Heating Units



Questions & Answers

Prof. Vince McDonnell
mcdonell@UCICL.uci.edu



Paul Glanville, PE
pglanville@gti.energy



Acknowledging our Project Partners and Funders:



Session #2 (3:00p-4:00p)

Research Project Plan & Results – Cont. (GTI)

- Review the experimental test plan and preliminary results for Industrial Combustion Equipment, Boilers, and Water Heaters

Hydrogen – OEM Perspective (A.O. Smith)

- Manufacturer perspective on H₂ applied to heating equipment and testing results

Pulling it All Together – H₂ Big Picture (GTI)

- Putting research data into broader context, including techno-economics, codes & standards, trends in test data, and H₂ safety